

Alkali Antimonide Photocathode Engineering

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Overview

A little bit about accelerators

Brief intro to 3-step model as it applies to semiconductors

Where do we need to be? What is limiting us now?

In situ materials analysis during cathode formation

How to grow smoother cathodes (and why you might want to)

Toward single crystals and heterojunctions

Reference Material

Some of this talk comes from a course on Cathode Physics Matt Poelker and I taught at the US Particle Accelerator School

<http://uspas.fnal.gov/materials/12UTA/UTA-Cathode.shtml>

https://science.energy.gov/~media/bes/pdf/reports/2017/Future_Electron_Source_Worskhop_Report.pdf

Modern Theory and Applications of Photocathodes

W.E. Spicer & A. Herrera-Gómez

SAC-PUB-6306 (1993)

Great Surface Science Resource:

<http://www.philiphofmann.net/surflec3/index.html>

Does the particle source matter?... Sometimes

- The electron beam properties determine the photon beam properties
 - Pulse duration, degree of coherence, flux
- In all light sources through 3rd generation, the phase space is determined by the ring



- In X-ray free electron lasers (LCLS II, XFEL, MaRIE), this will change – the electron source will determine the beam properties
- The highest brightness sources available are photoinjectors, which use a laser on a photocathode to control the spatial and temporal profile of the emitted electron beam

Applications at the State of the Art

Electron cooling of ion machines

Requires high current with long operational life, other requirements are modest (~ 50 mA with $5\mu\text{m}$ emittance)

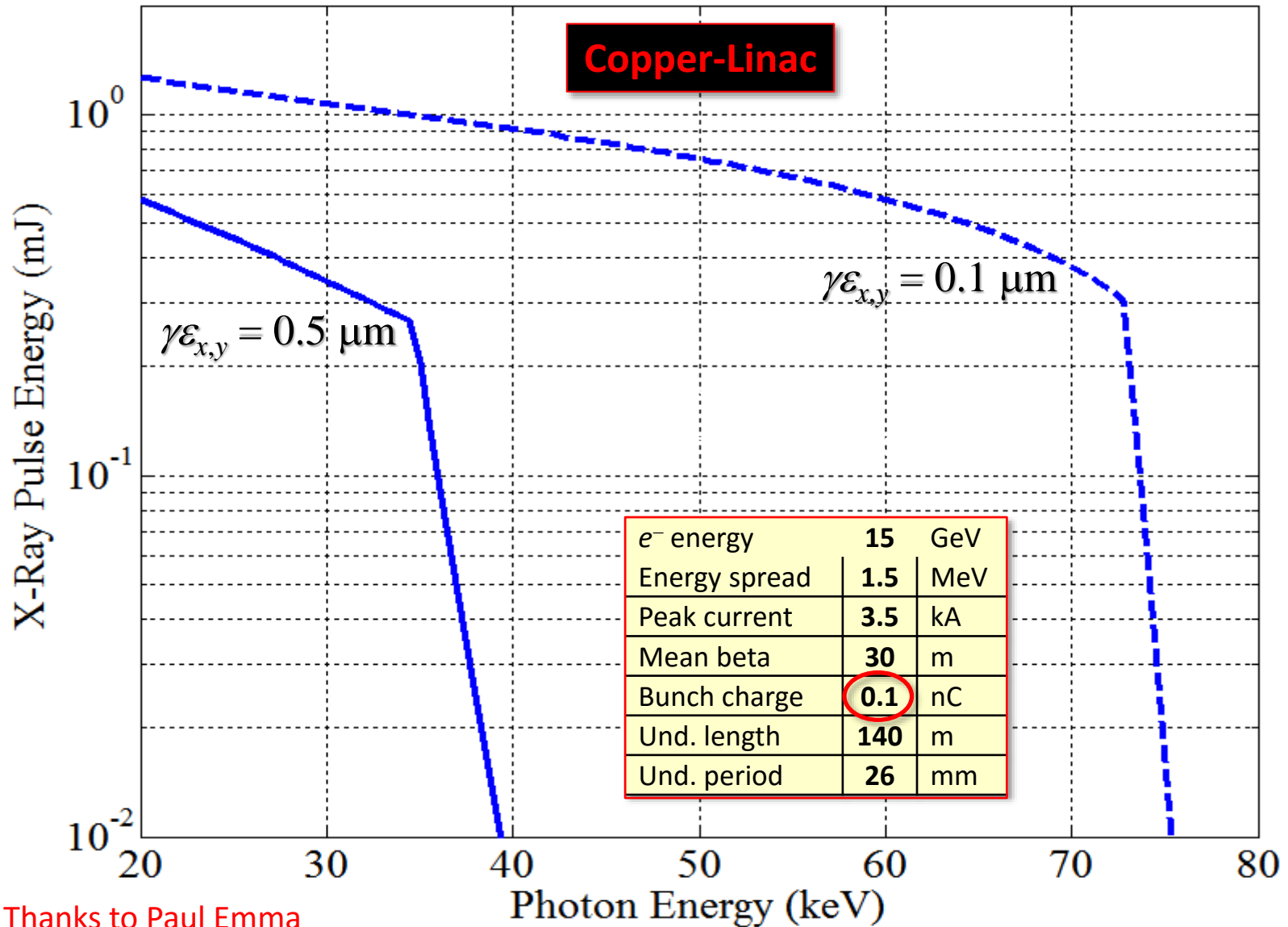
FEL sources

Going to moderate currents (still under 1 mA); emittance improvement is a big deal (ideally $0.1\ \mu\text{m}$)

Ultrafast Electron Diffraction/Microscopy

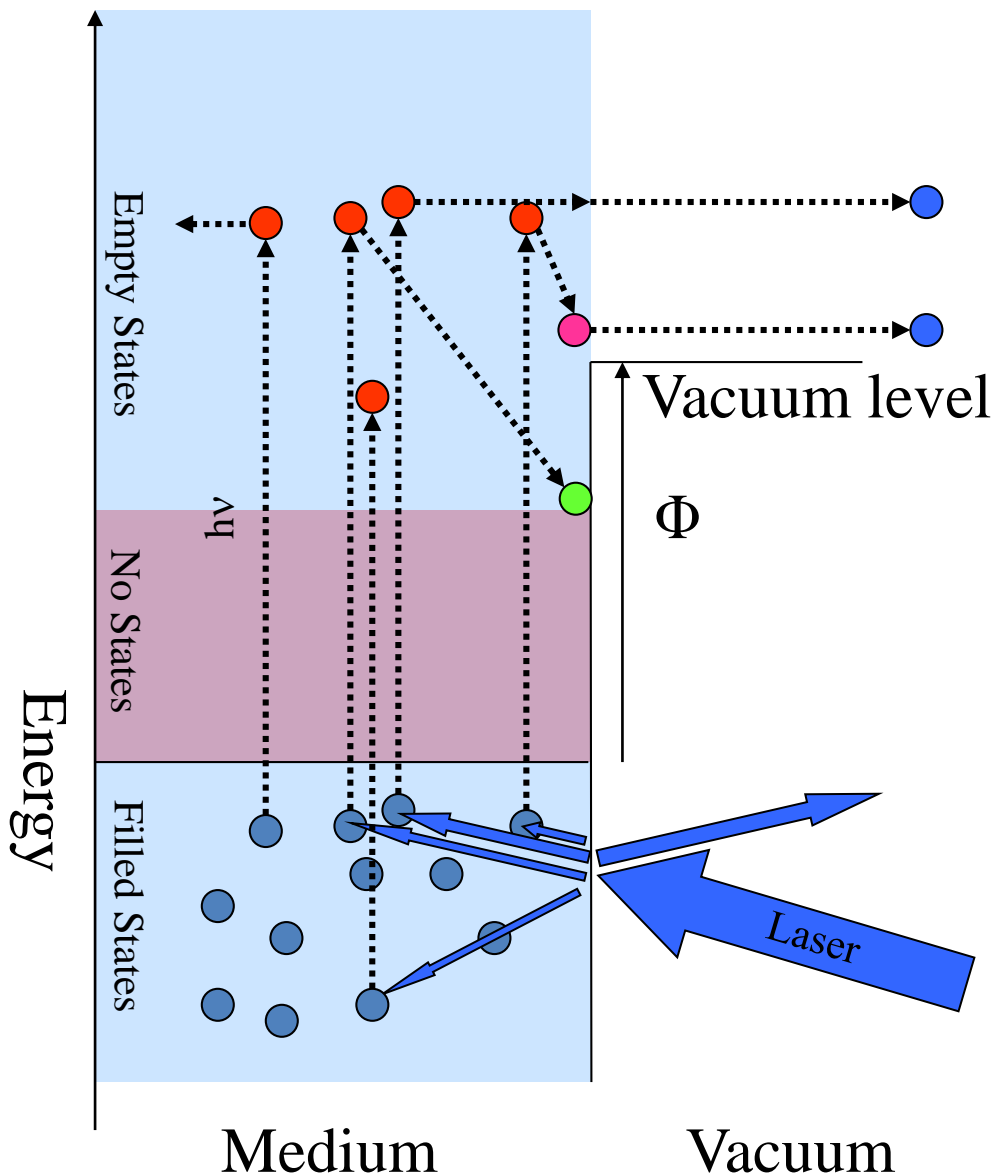
High brightness! Ideally a factor of 100 from current photoinjectors. Very low current. Short pulse duration (100 fs at sample, less for some applications)

Emittance Leverage at LCLS-II/HXR (15 GeV, 120 Hz)



Thanks to Paul Emma

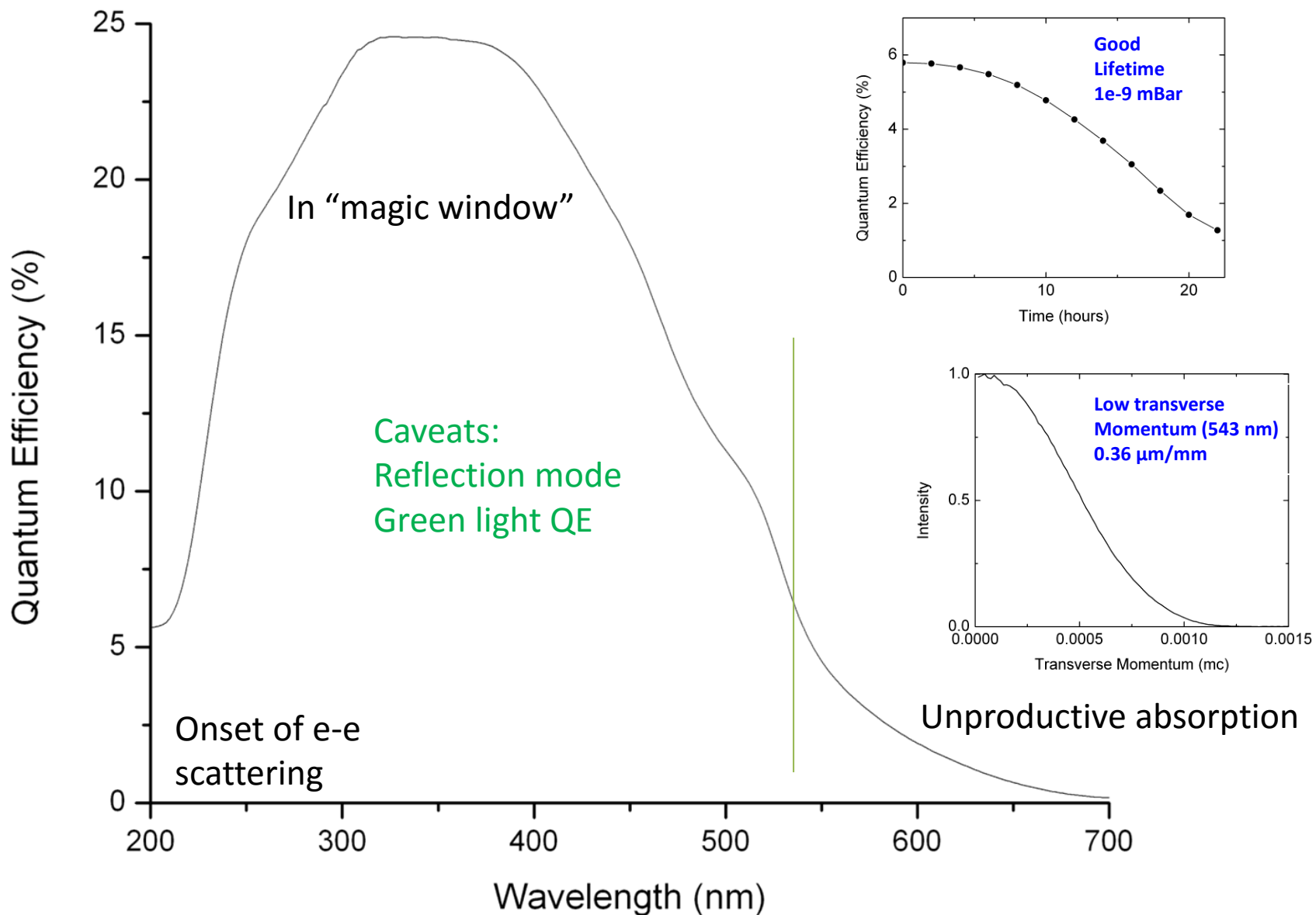
Three Step Model of Photoemission - Semiconductors



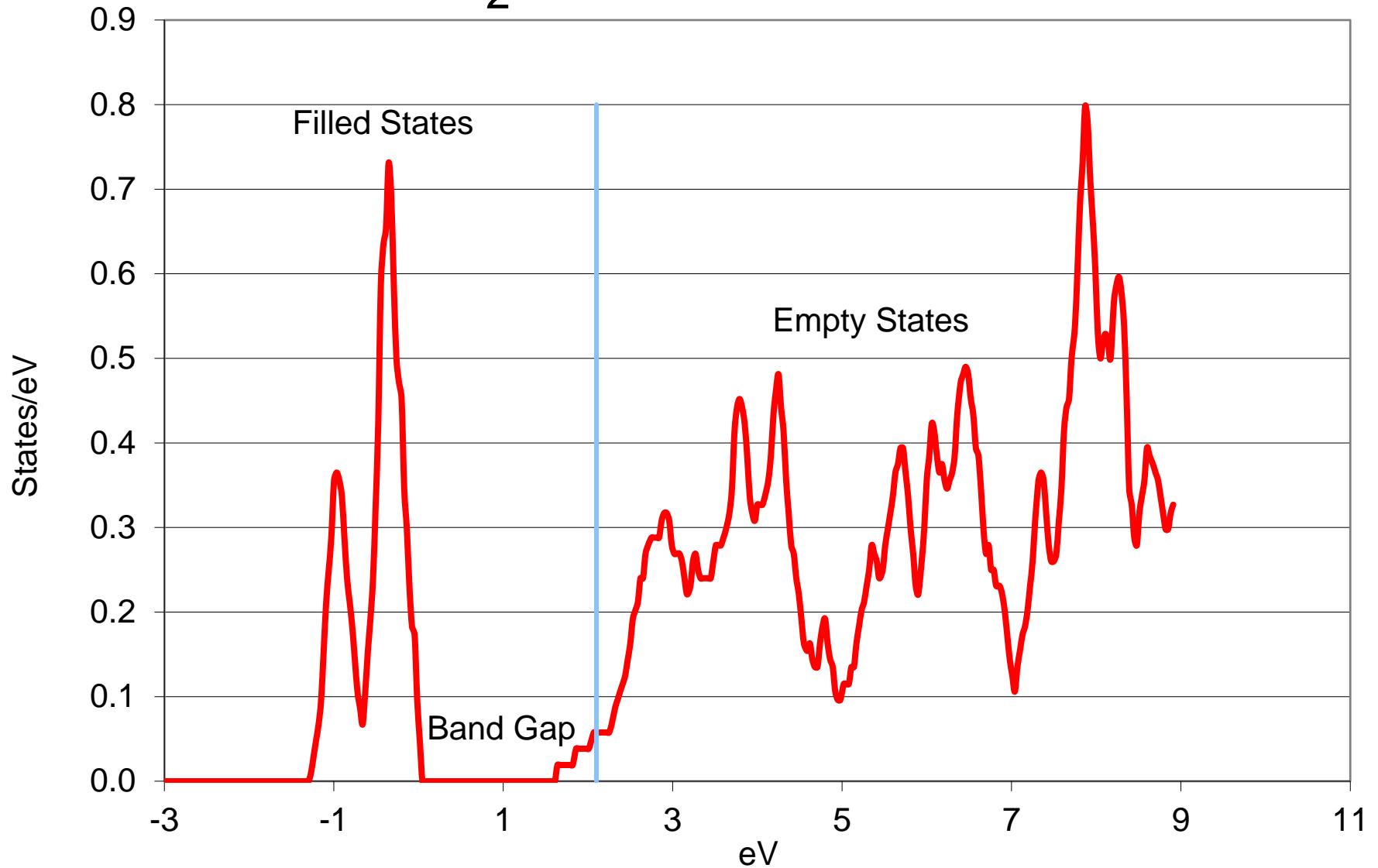
- 1) Excitation of e^-
Reflection, Transmission
Energy distribution of excited e^-
- 2) Transit to the Surface
 e^- -phonon scattering
 e^- -defect scattering
 e^-e^- scattering
Random Walk
- 3) Escape surface
Overcome Workfunction
Multiple tries

Need to account for Random Walk in cathode suggests Monte Carlo modeling

K₂CsSb: A Good Candidate

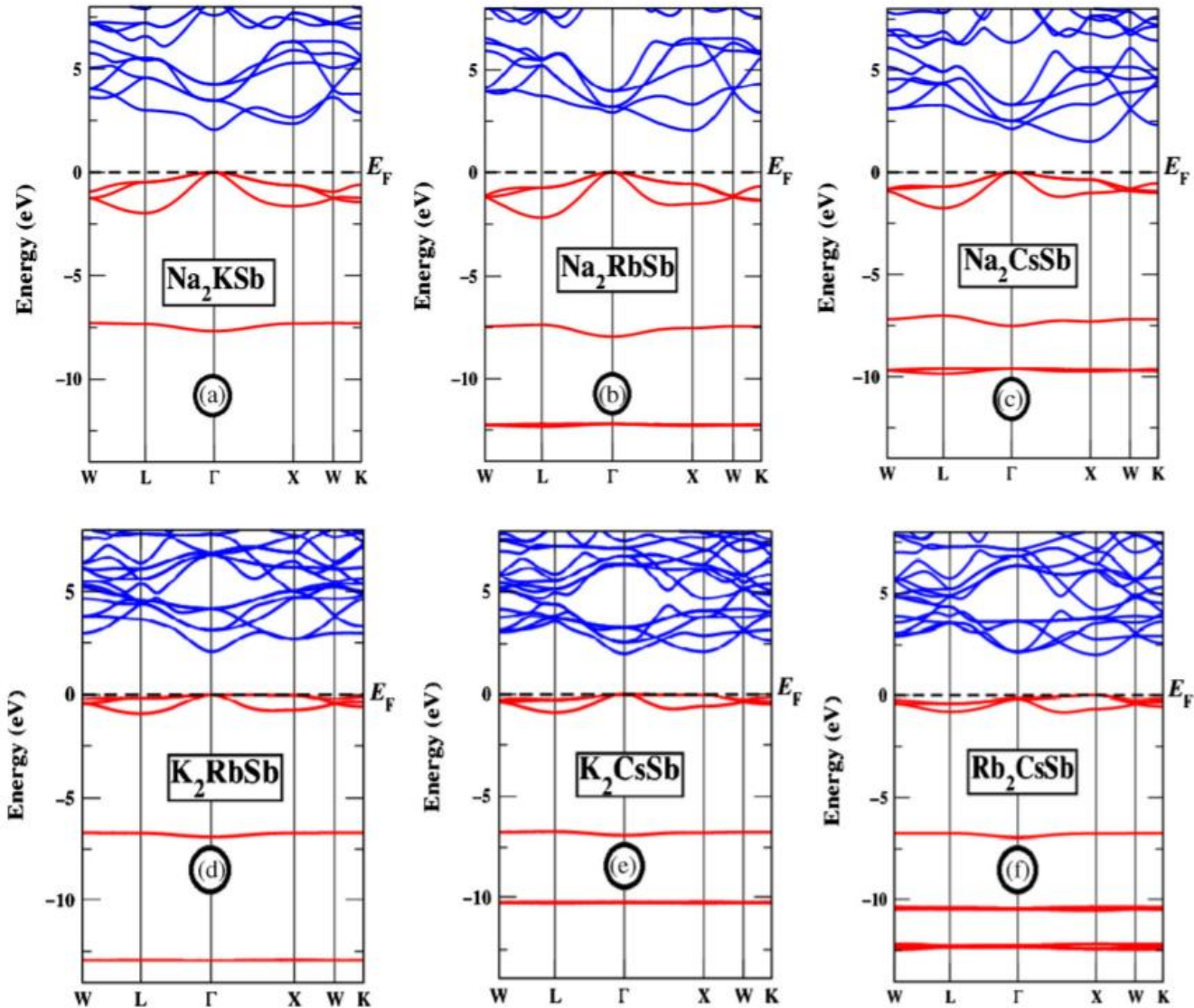


K_2CsSb DOS



A.R.H.F. Ettema and R.A. de Groot, Phys. Rev. B **66**, 115102 (2002)

Alkali Antimonide Family



Parameters, and how to affect them

Increasing the electron MFP will improve the QE. Phonon scattering cannot be removed, but a more perfect crystal can reduce defect and impurity scattering:

$$\frac{1}{\lambda_{MFP}} = \frac{1}{\lambda_{el-el}} + \frac{1}{\lambda_{ap}} + \frac{1}{\lambda_{op,ems}} + \frac{1}{\lambda_{op,abs}} + \frac{1}{\lambda_{impurity}} + \frac{1}{\lambda_{defect}} + \frac{1}{\lambda_{boundary}}$$

Control of surface roughness is critical to minimizing the intrinsic emittance – epitaxial growth?

A question to consider: Why can CsI (another ionic crystal, PEA cathode) achieve QE>80%?

T.H. Di Stefano and W.E. Spicer, Phys. Rev. B **7**, 1554 (1973)

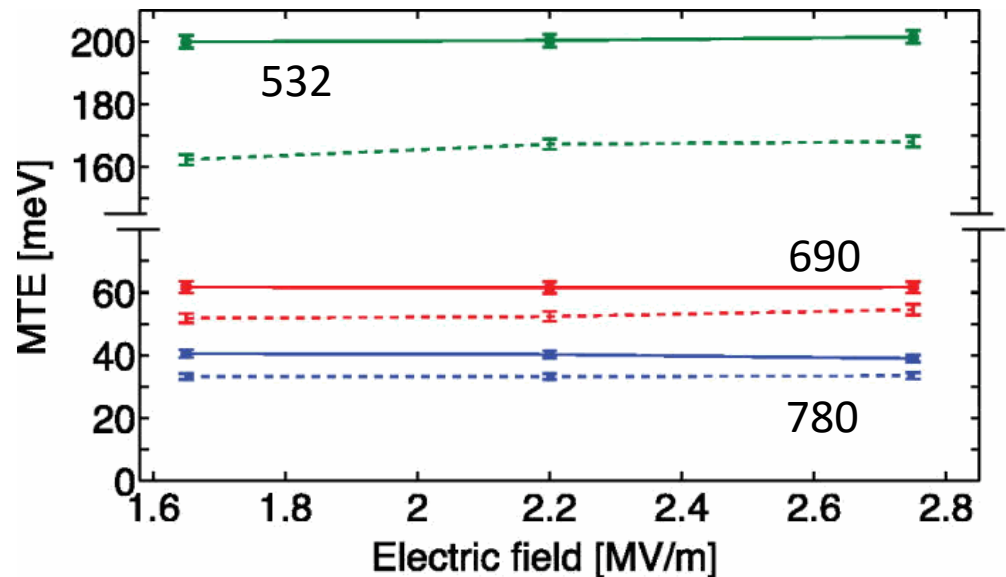
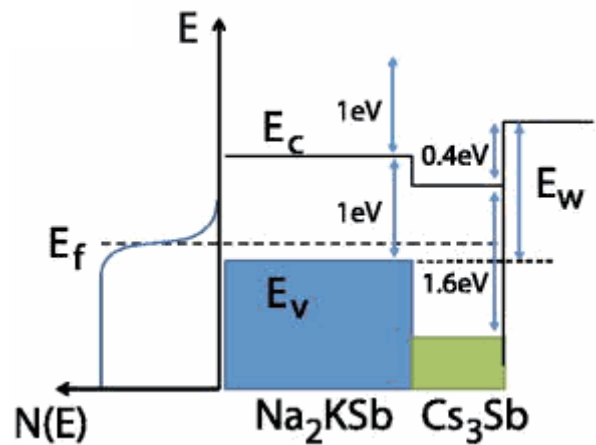
Large band gap and small electron affinity play a role, but, so does crystal quality.

Scattering

Phonon scattering can be helpful

Well known in thermalization of electrons in GaAs and Diamond

Luca & company recently demonstrated in PEA materials:
20% reduction in MTE for transmission mode operation



150 nm Na_2KSb (with few nm surface layer)

Appl. Phys. Lett. **108**, 124105 (2016)

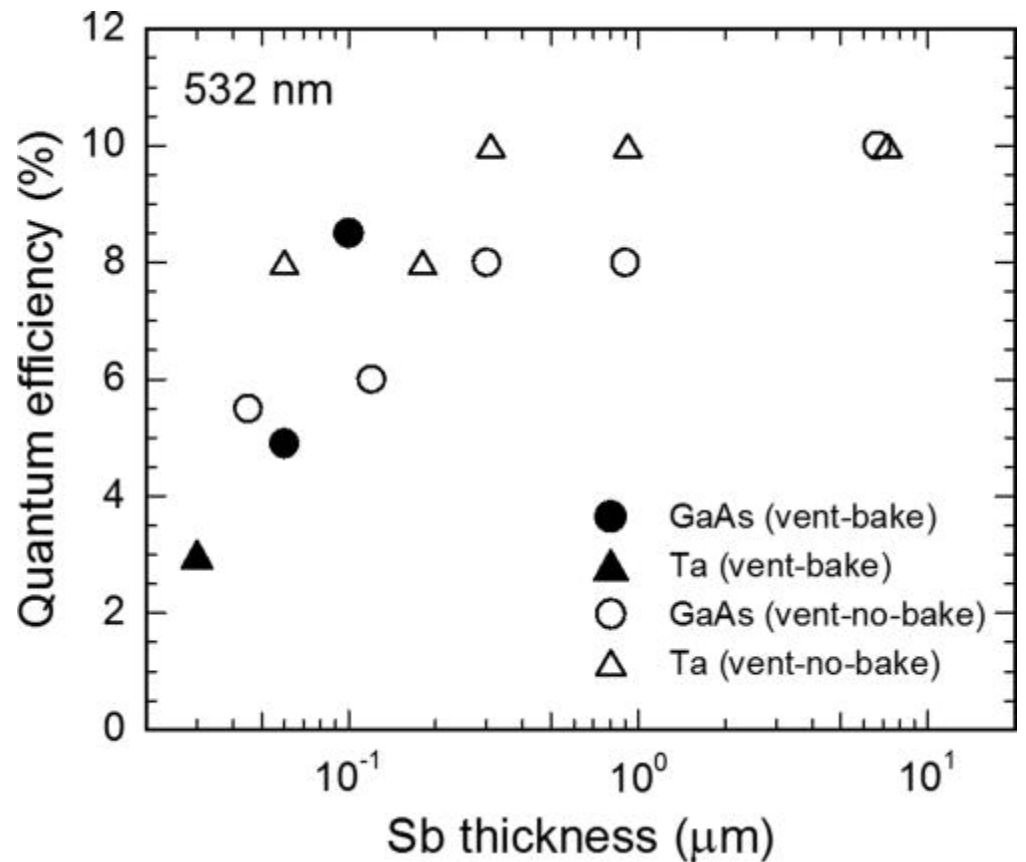
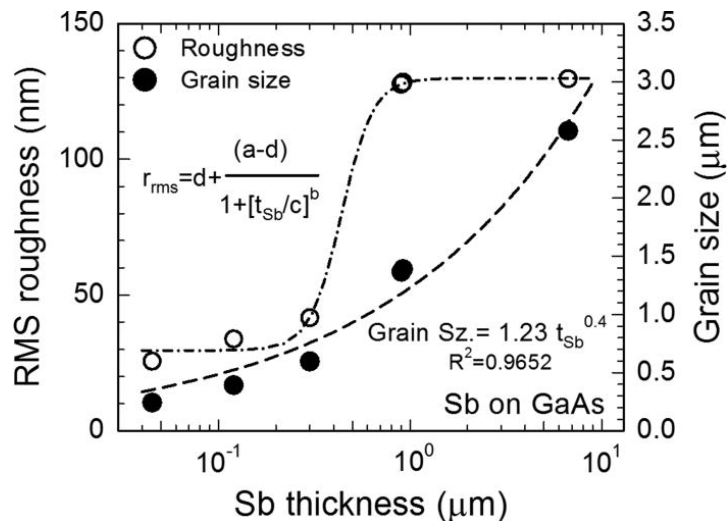
Grain Boundaries

Grain boundary scattering: just bad

Large grains or Epitaxy

One solution: REALLY thick cathodes (50 optical absorption lengths)

Another: Stay tuned



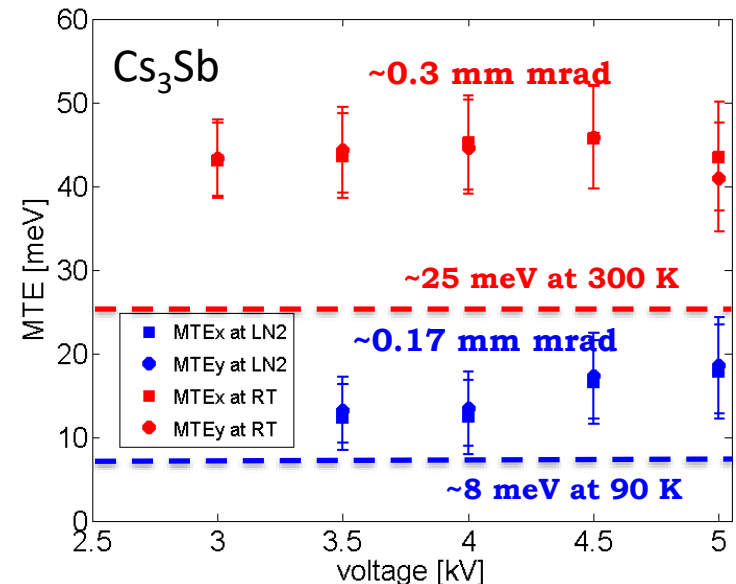
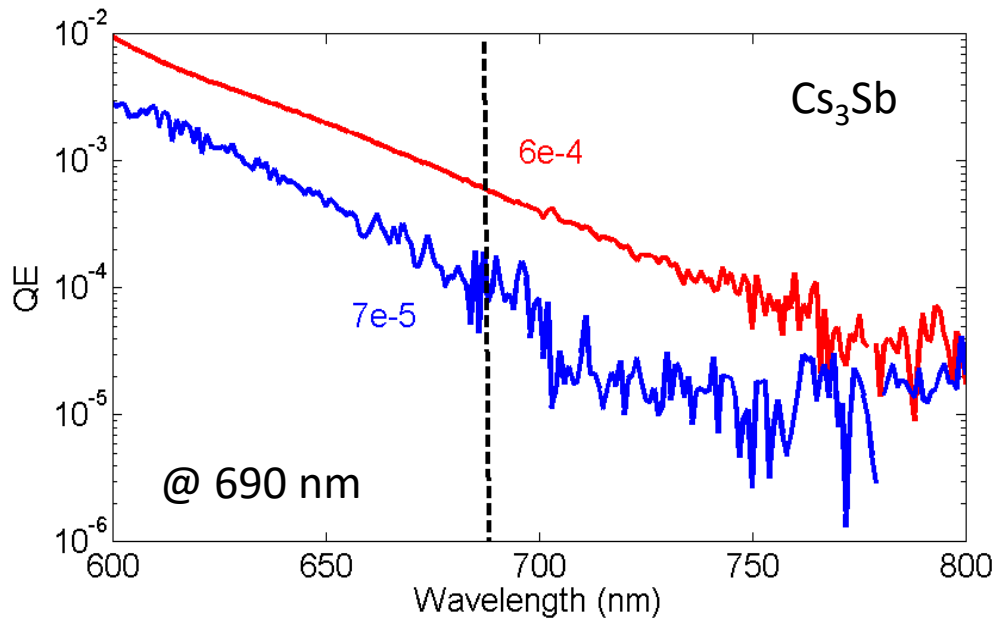
Lattice Temperature

For $E_{\text{excess}} < 0$, $MTE = kT$, and lattice temperature becomes important

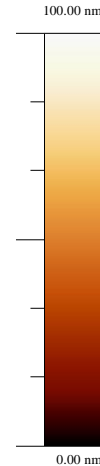
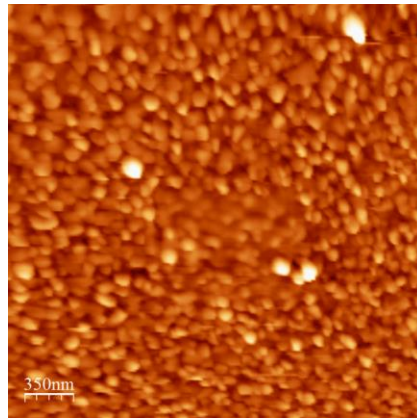
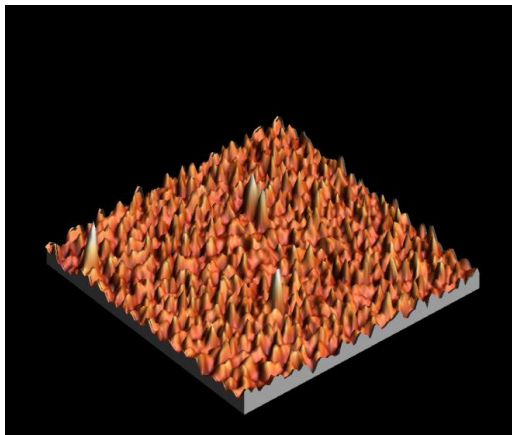
Photoemission in this domain relies on defect states

May depend on crystal quality

Material conductivity will as well



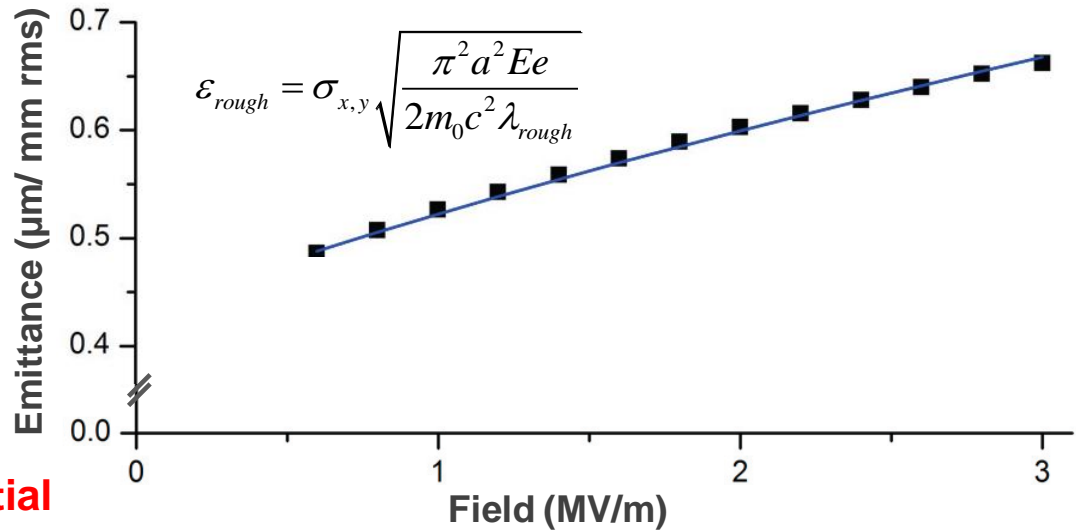
Roughness and Emittance



**25 nm roughness,
100 nm spatial period**

S. Schubert et al., APL Materials 1, 032119 (2013)

**Emittance vs field
measured with
Momentatron, 532 nm light**



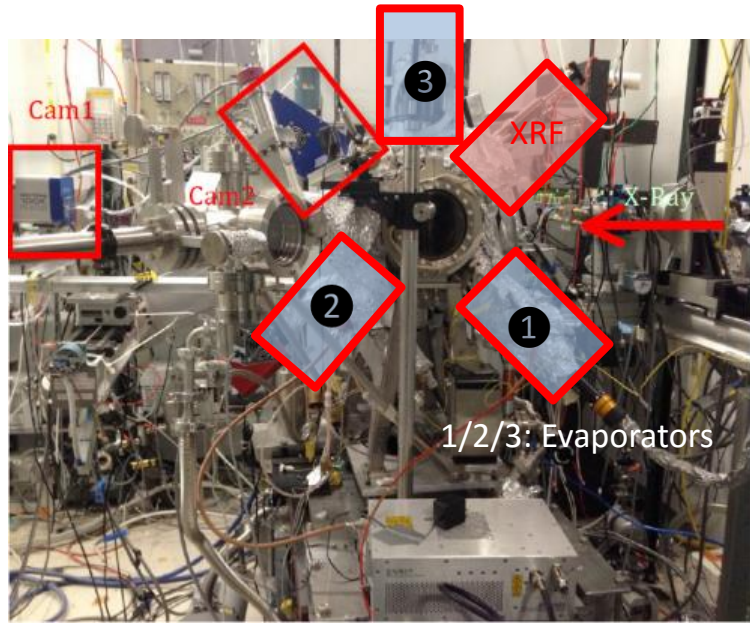
T. Vecchione, et al, Proc. of IPAC12, 655 (2012)

**We now understand why sequential
growth causes roughness, and can
achieve near-atomic roughness with
Alkali antimonides!
(Next Section)**

In operando analysis during growth

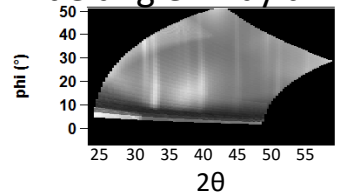
(setup at NSLS/X21 & CHESS G3 – ISR soon)

- Growth and characterization system

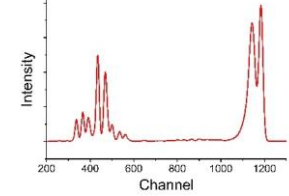


In-situ UHV growth system ($10^{-10} \sim 10^{-11}$ Torr) installed at G3, CHESS

Wide angle X-ray diffraction



X-ray fluorescence

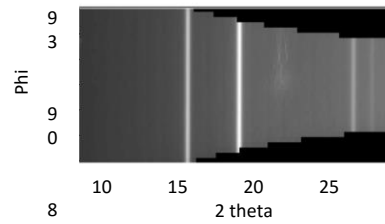


Camera 2 (PILATUS 100K): WAXD

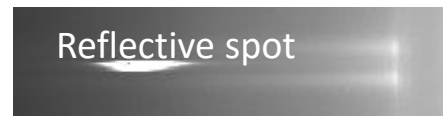
Fluorescence detector (Vortex)

Synchrotron Radiation
+ : high brilliance, ...
- : limited beamtime

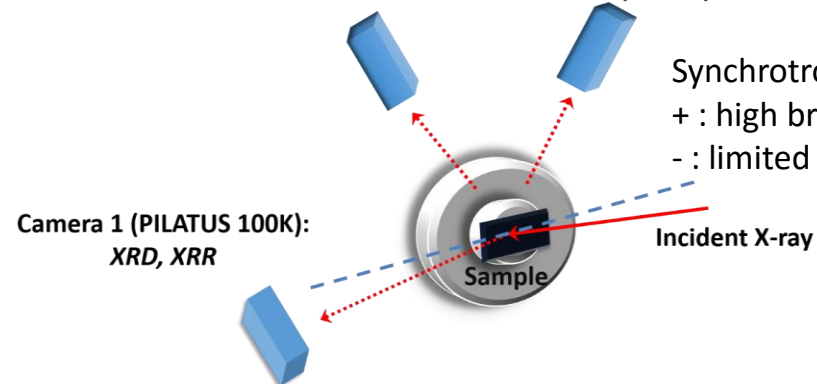
Camera 1 (PILATUS 100K): XRD, XRR



X-ray diffraction

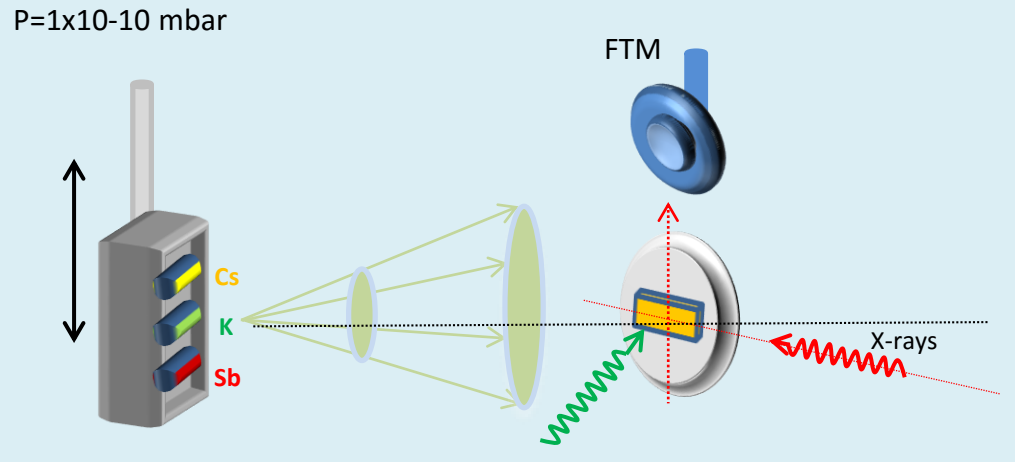


X-ray reflectivity

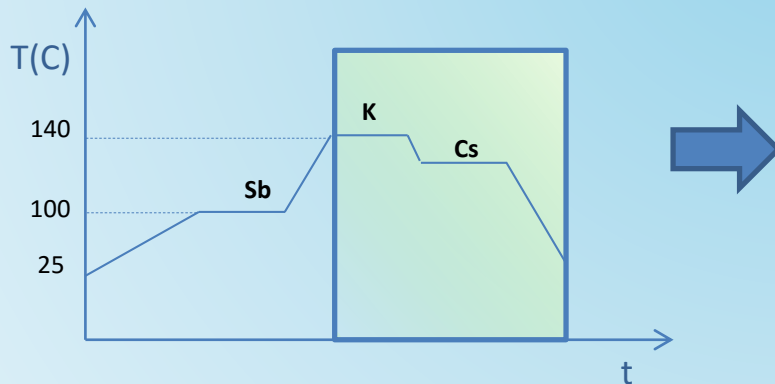


Experimental set up: K_2CsSb cathode growth

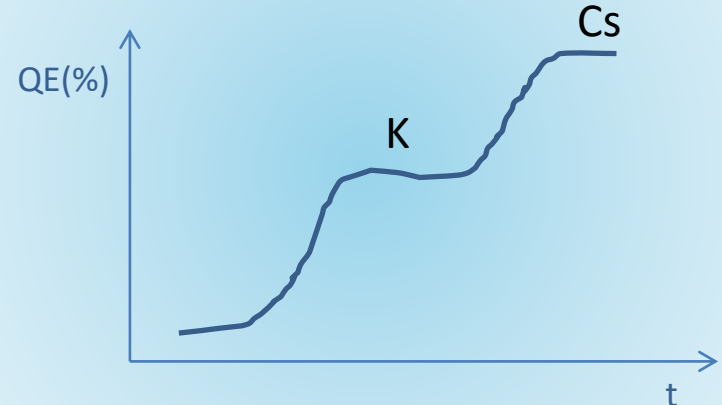
Horizontal evaporation of three sources:



Recipe:



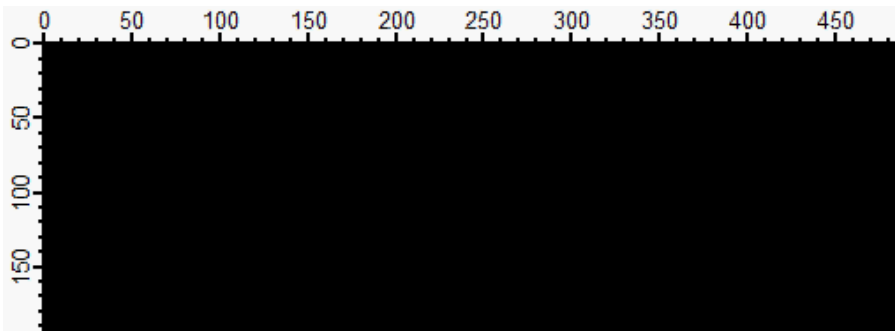
QE during growth (532 nm laser)



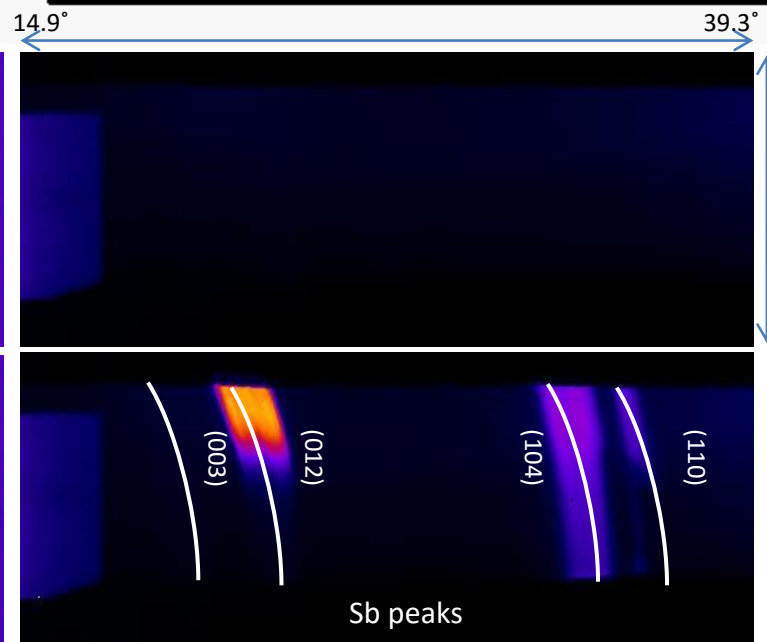
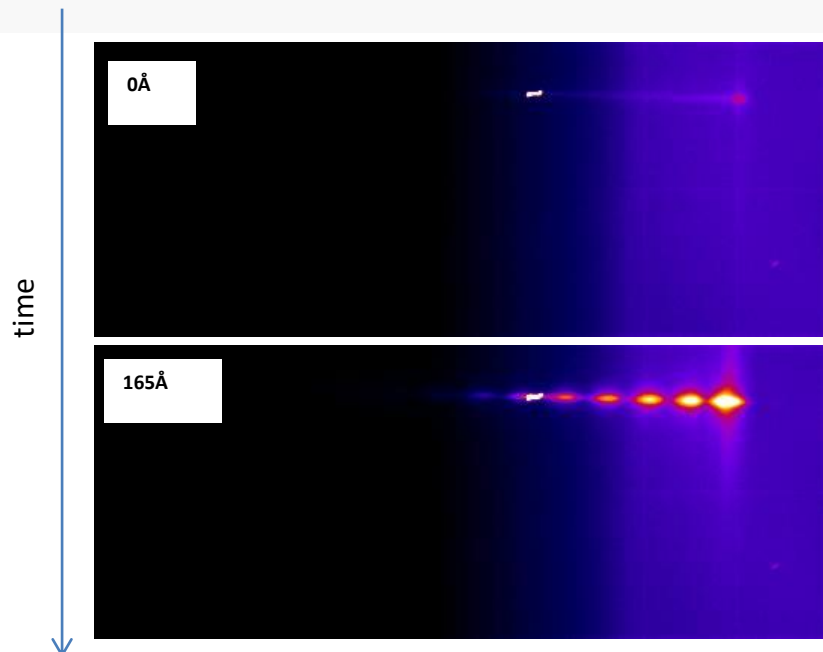
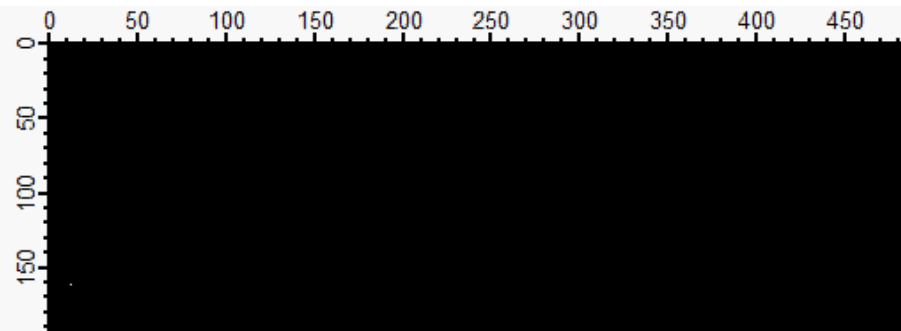
Simultaneously Acquire XRD and GISAXS

- Understanding reaction dynamics through crystalline phase evolution
- Map the thickness and roughness evolution of the cathode
- Is there a correlation between reactivity, QE and roughness?

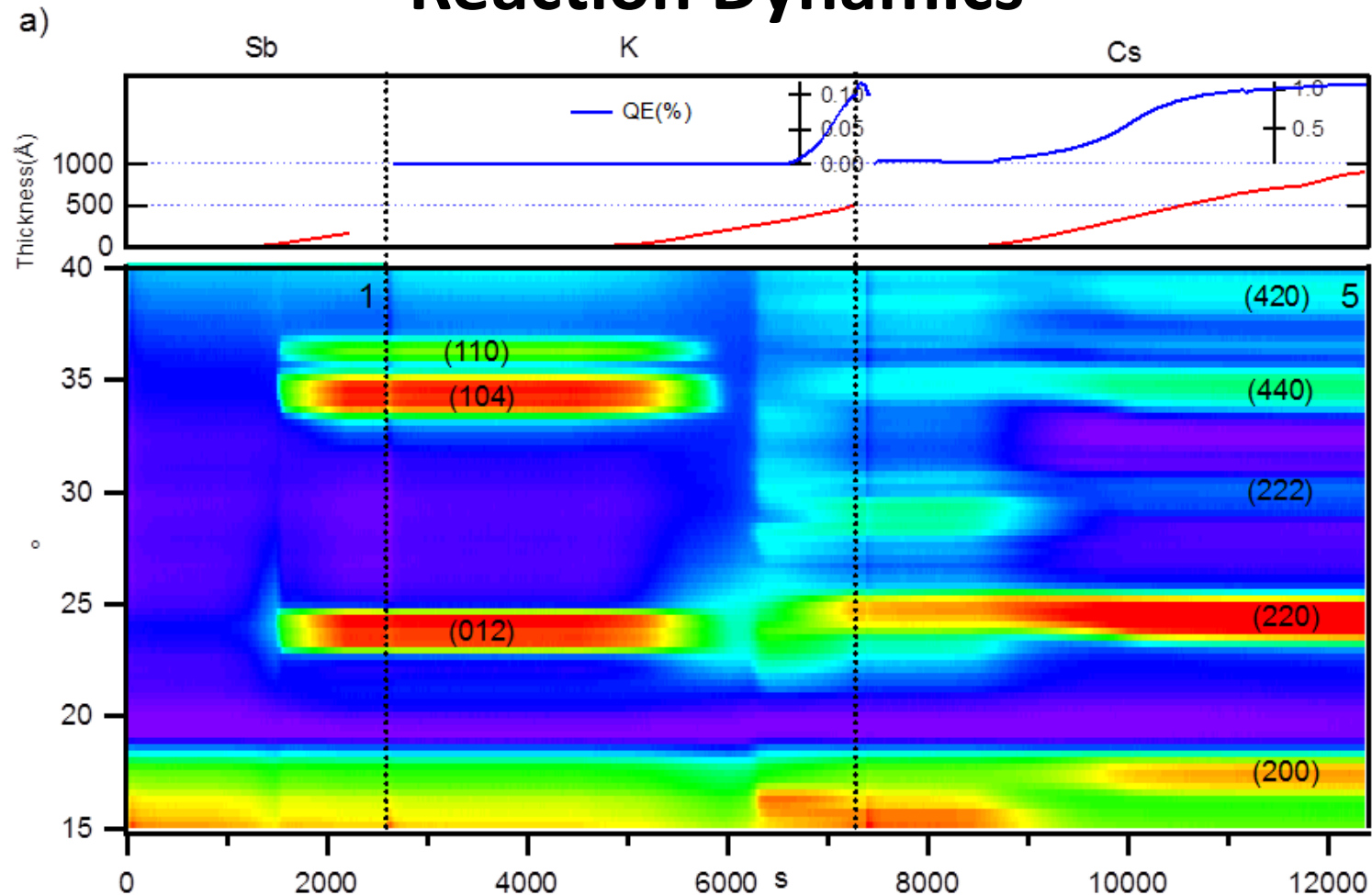
Camera 1: GISAXS & XRR



Camera 2: WAXS



Reaction Dynamics



b)

Antimony evaporated on Si, 0.2 \AA/s ; crystallize at 4nm
K deposition dissolves Sb layer - This is where roughening occurs!

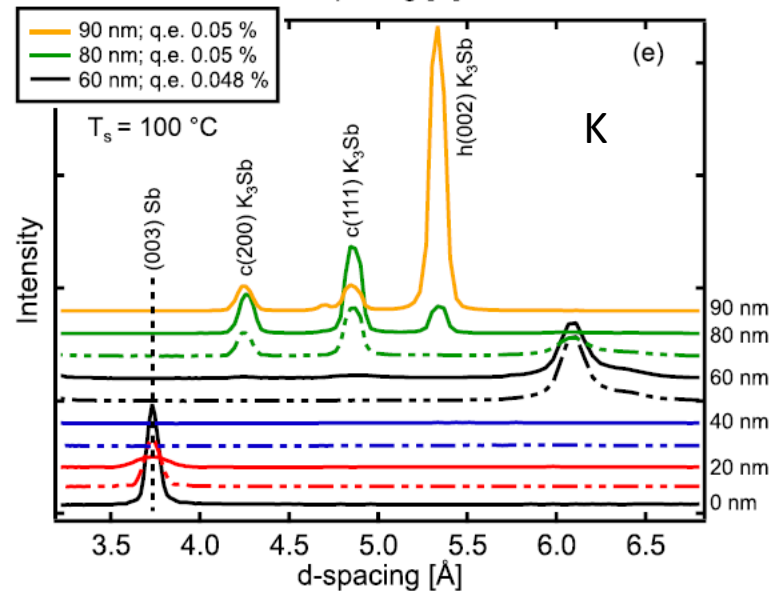
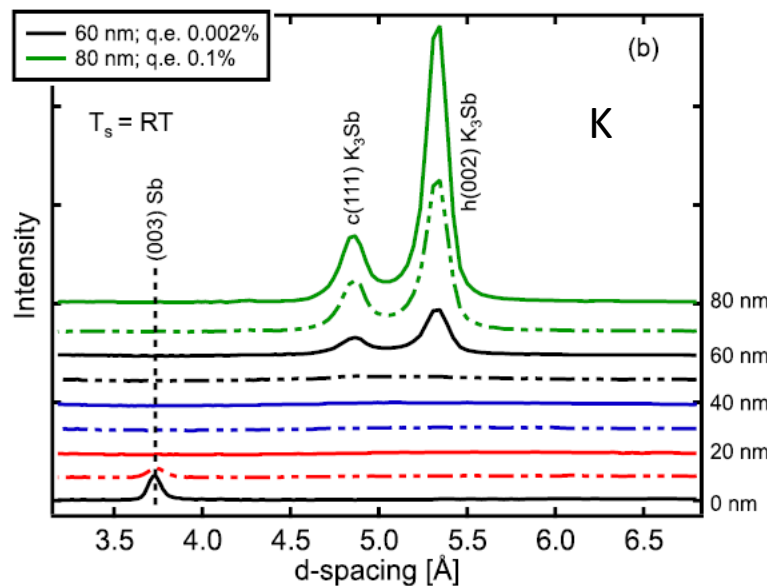
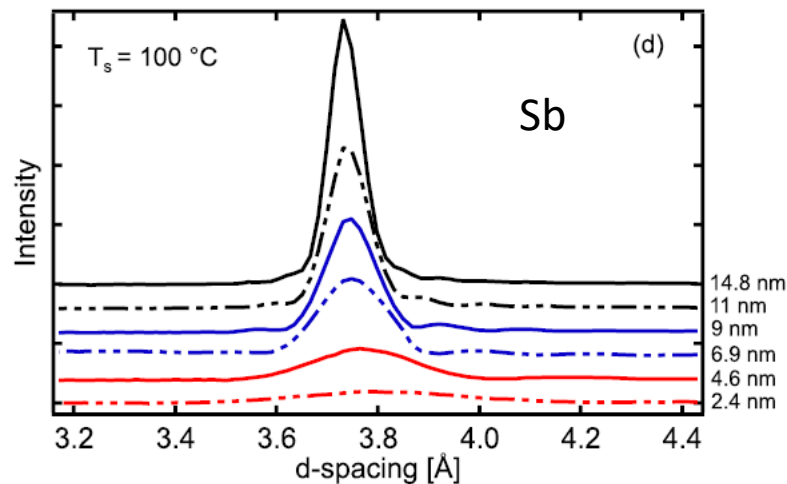
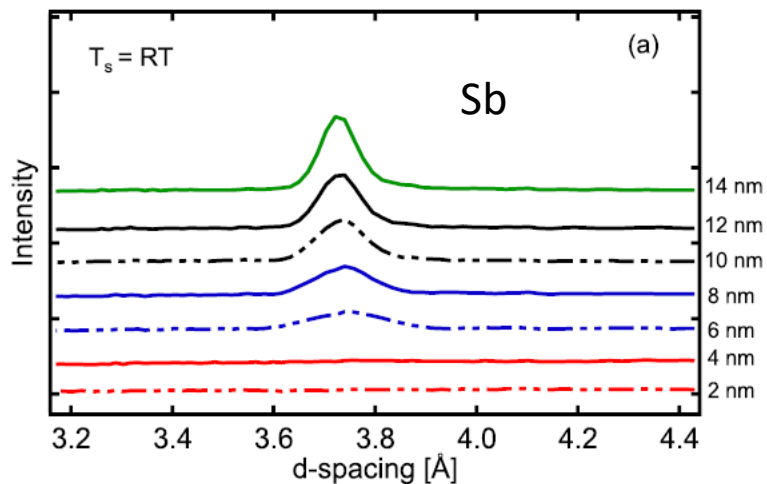
QE increase corresponds with $K_x\text{Sb}$ crystallization

Cs increases lattice constant and reduces defects

M. Ruiz-Osés et al., APL Mat. 2, 121101 (2014)

Stepwise High Resolution XRD

A little bit of Potassium goes a long way...



Stepwise High Resolution XRD

A little bit of Potassium goes a long way...

Room Temperature

100C Substrate

Recrystallization to K_3Sb

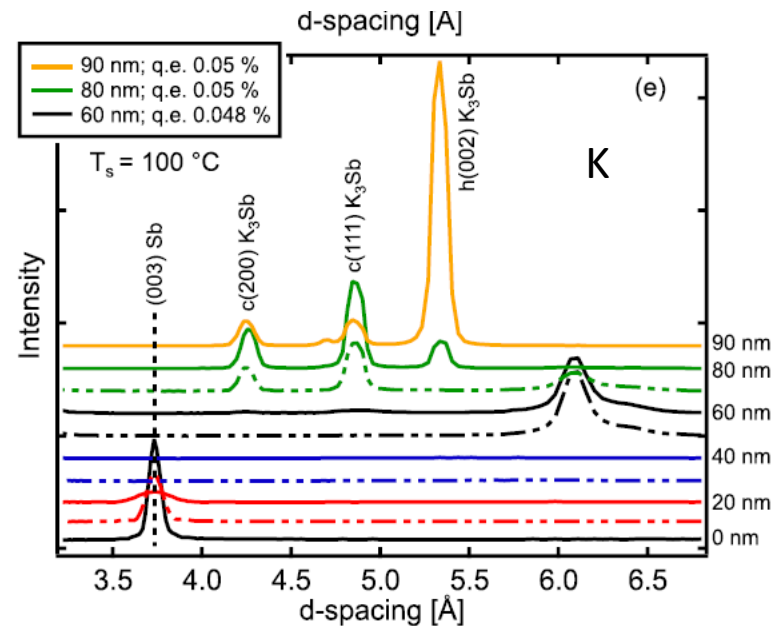
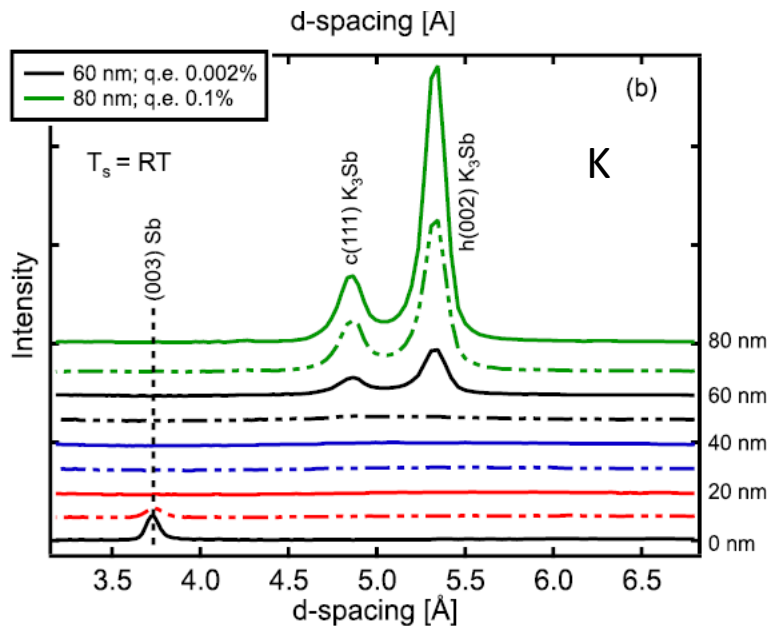
Recrystallization to K_xSb

Better QE of K_3Sb

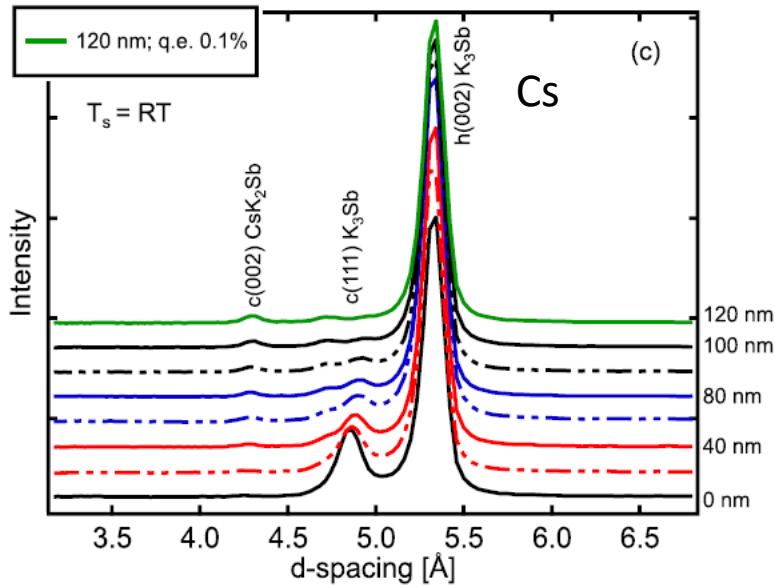
Cubic K_3Sb first

Principally Hexagonal

Eventually Hex appears (oops!)

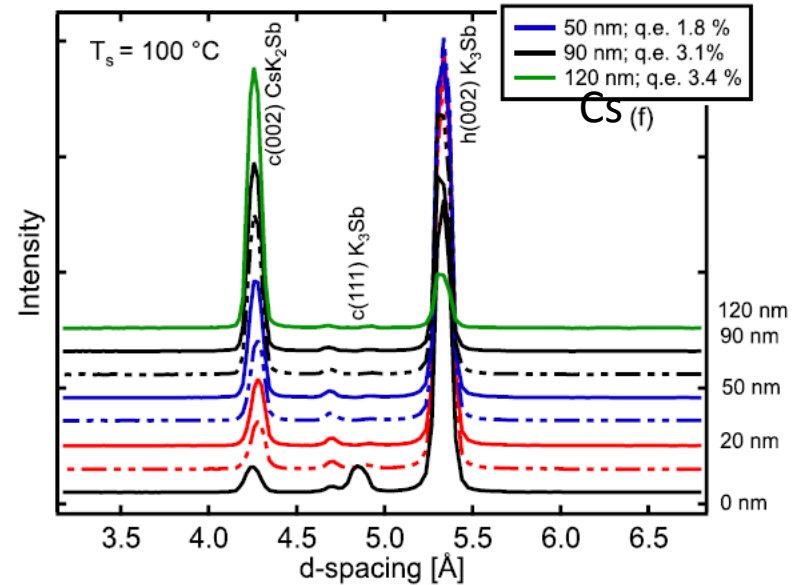


Stepwise High Resolution XRD



Room Temperature

K₃Sb resists Cs incorporation
QE never improves



100C Substrate

Cubic K₃Sb converts quickly
Hex K₃Sb mostly converts

Stepwise High Resolution XRD

100C without “too much” K

15 nm Sb, 70 nm K

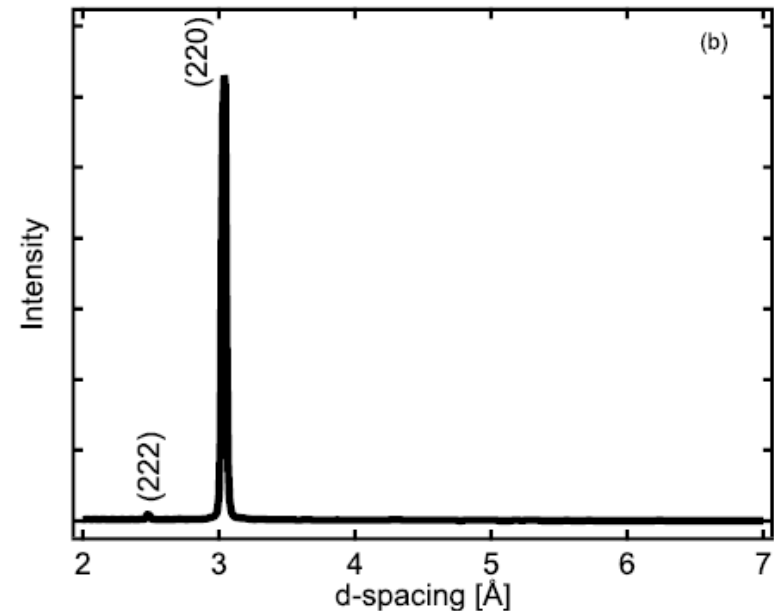
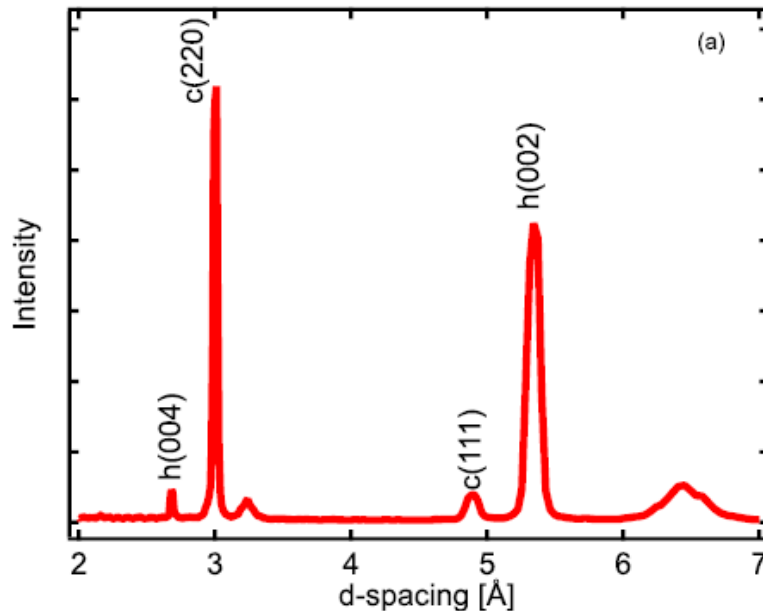
Stop at “mixed phase” K_xSb

lower QE of K_3Sb

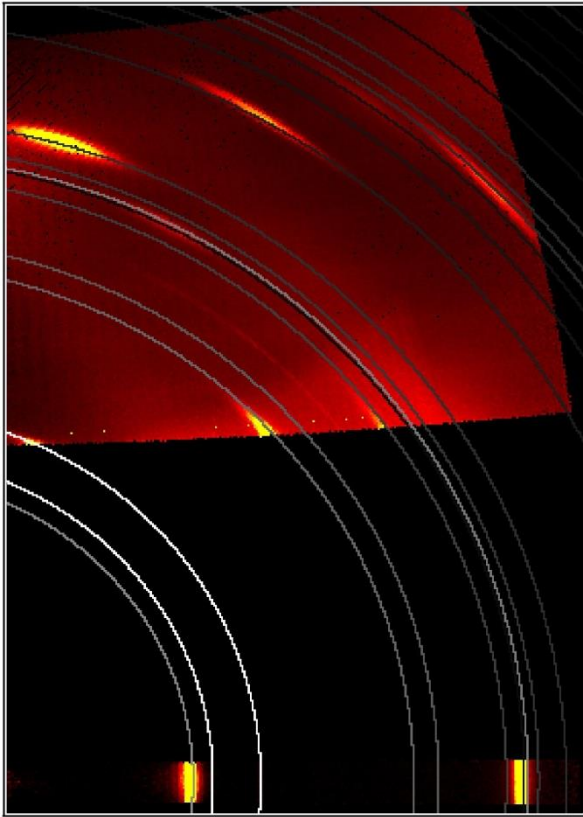
90 nm Cs sufficient

Full conversion to CsK_2Sb

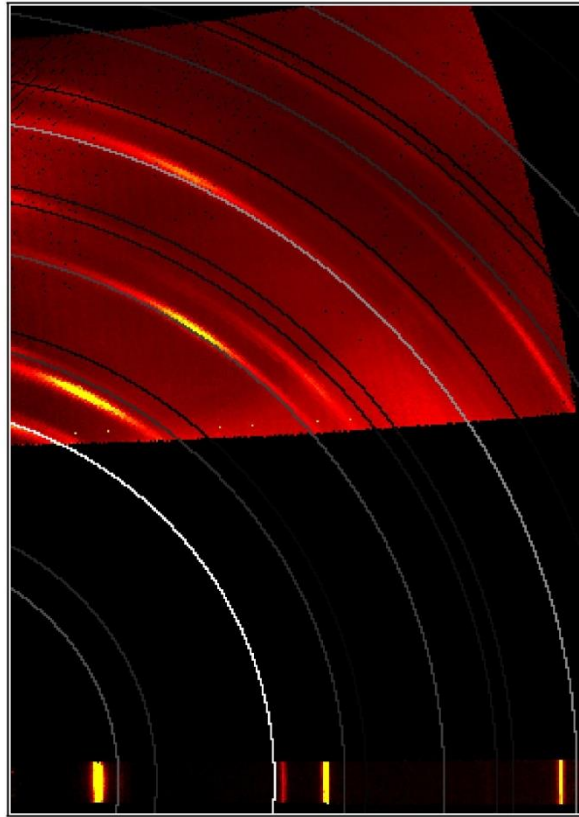
QE = 6.7% at 532 nm



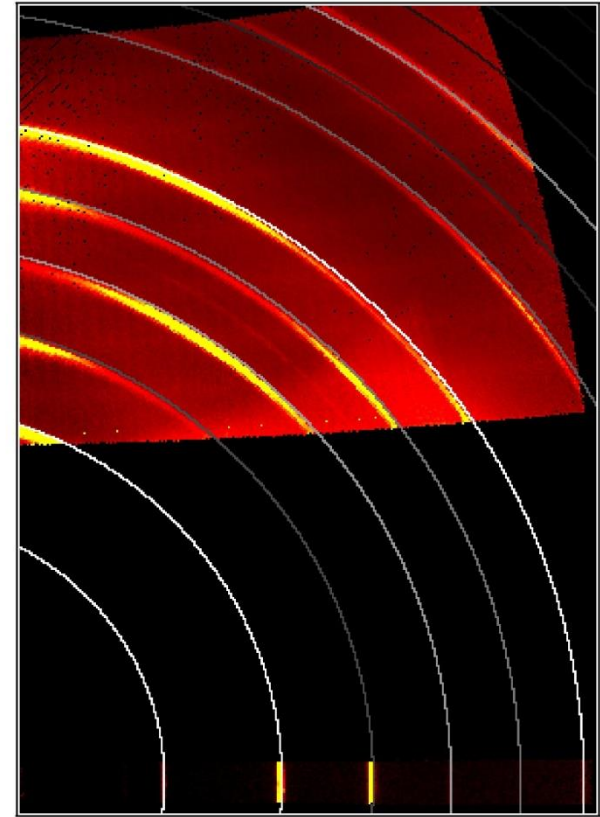
Cathode Texture



Sb evaporated at RT
Clear [003] texture



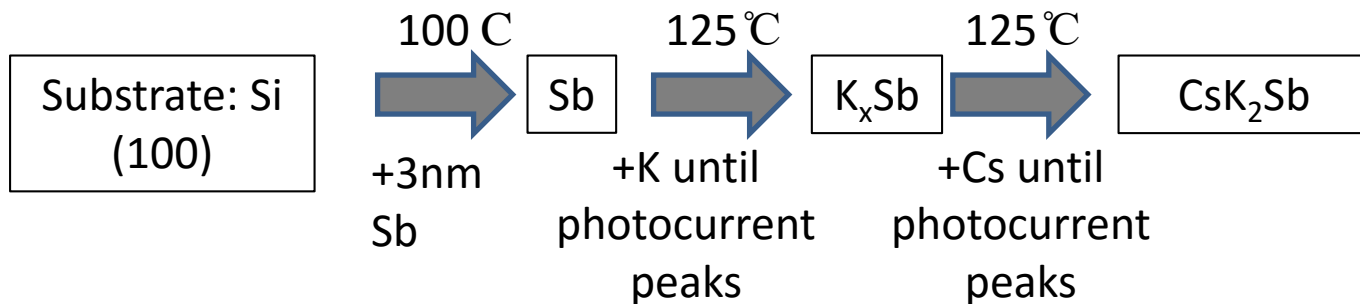
Add Potassium at 140C
Textured final film
But not K_3Sb



Add Cesium at 140C
Textured final film
Both [220] & [222]
(domains?)
Final QE 7.5% @ 532nm

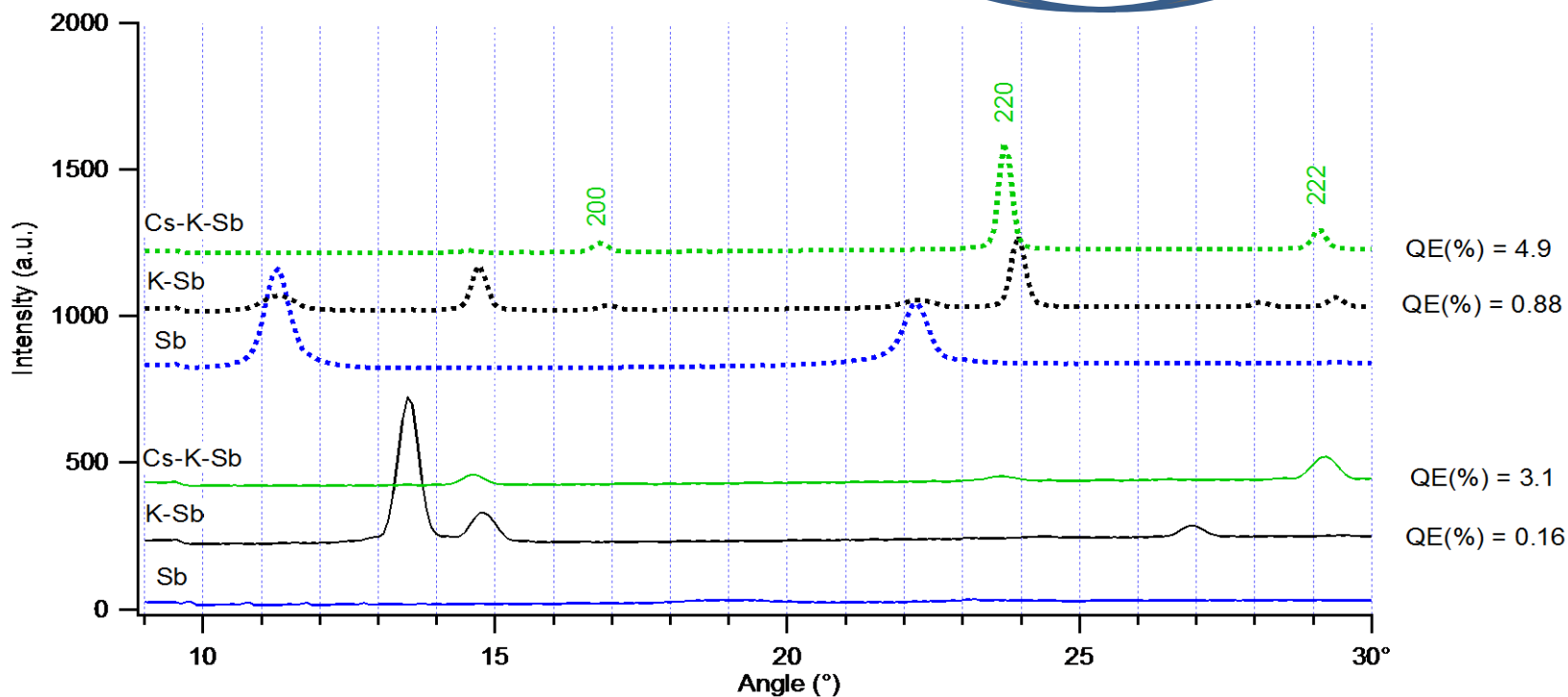
Engineering a Smoother Cathode

Idea: Never let Sb crystallize

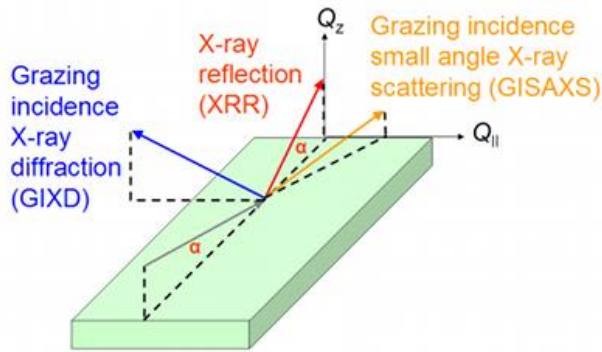


M. Ruiz-Osés et al., APL Mat. 2, 121101 (2014)

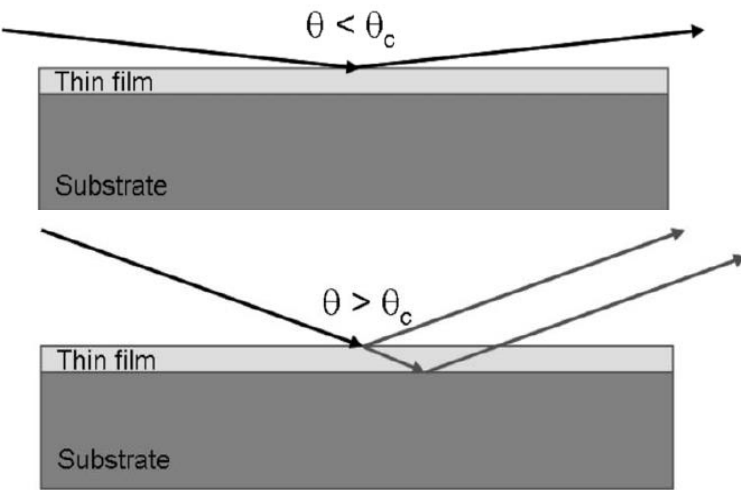
2nd layer: +5nm Sb



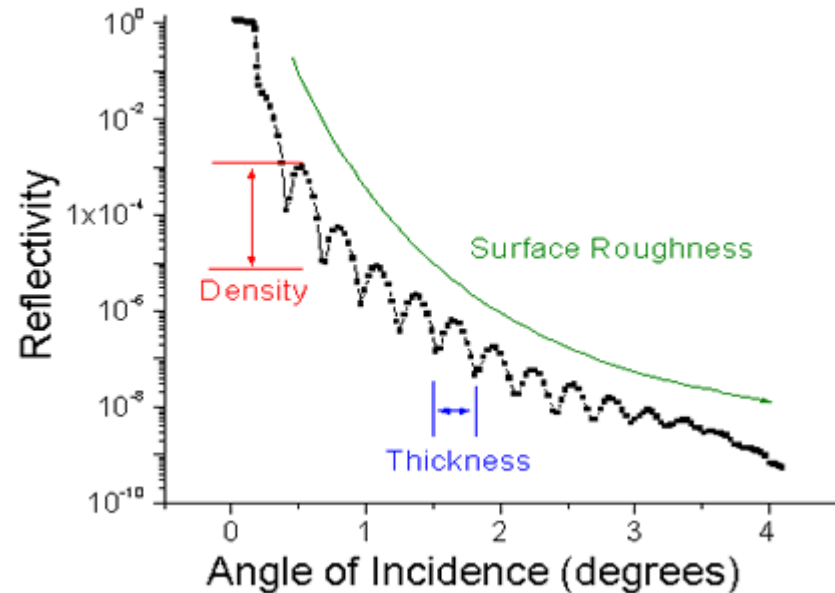
X-ray reflectometry (XRR) provides in-situ thickness monitoring



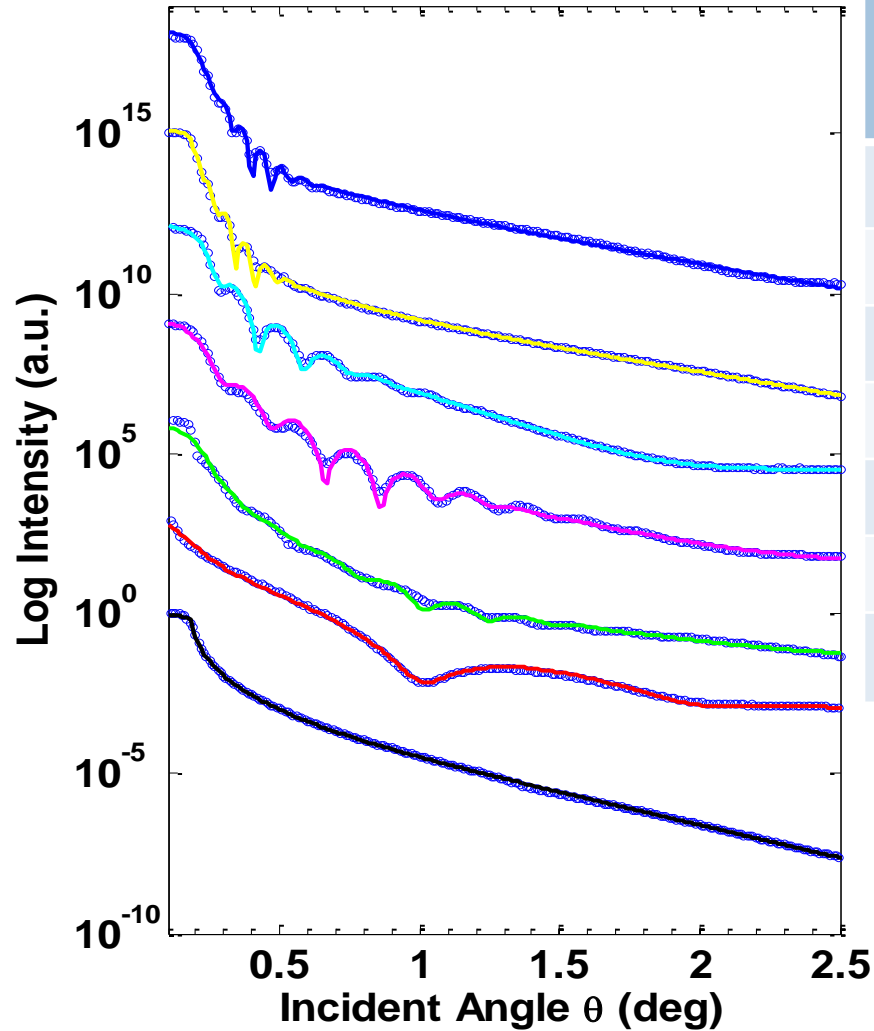
$$\theta_c = \arccos(n_{medium} / n_{air})$$



- Understand 'sticking' coefficient of materials to substrates at various temperatures
- Observe the intermixing vs layering of materials
- Observe the onset of roughness



XRR shows roughness evolution

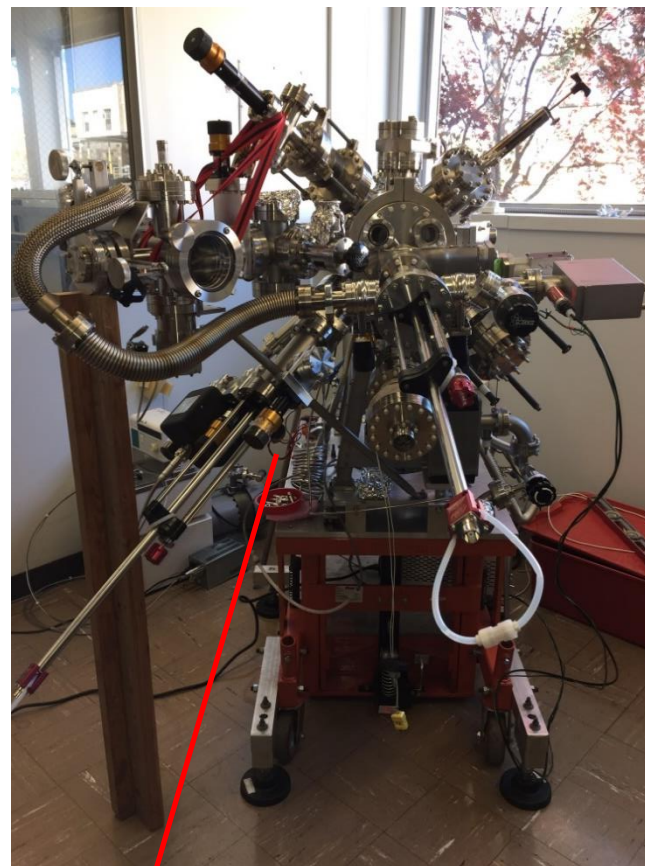
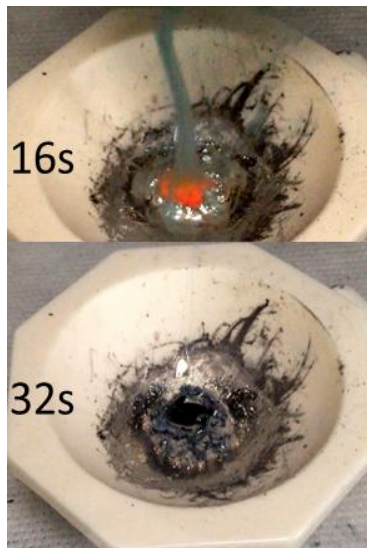


Deposited Layers	Total Thickness (Å)	Roughness (Å)
Cs-K-Sb-Cs-K-Sb/Si	469	32
K-Sb-Cs-K-Sb/Si	449	36
Sb-Cs-K-Sb/Si	200	21.3
Cs-K-Sb/Si	174	13.2
K-Sb/Si	141	10.5
Sb/Si	35	2.9
Si Substrate	-	3.1

The substrate fit includes 1.5 nm of SiO_2

Multi-layer subcrystalline film is smoother,
At slight loss of QE

Sputter growth of Bi-alkali photocathode



K_2CsSb
sputter gun



Before



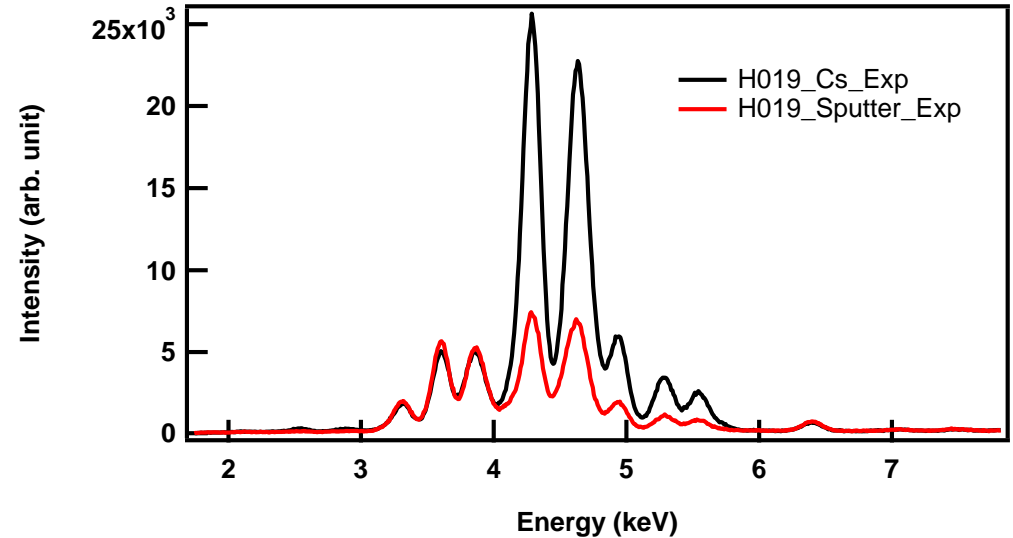
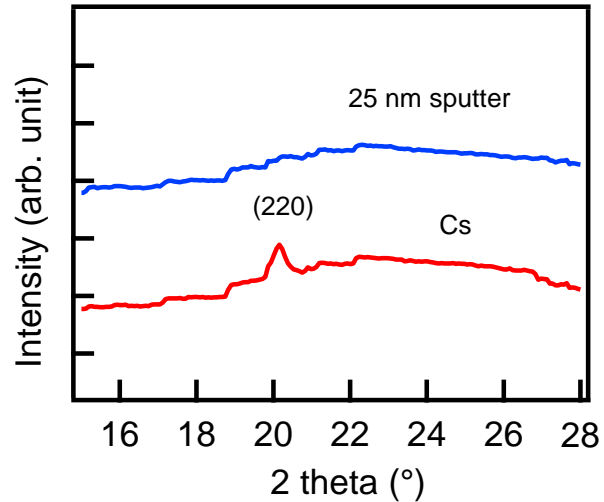
Sputtering

Sputter target and sputter gun was product of RMD. Inc. Photos of sputter target prep are contributed by H. Bhandari

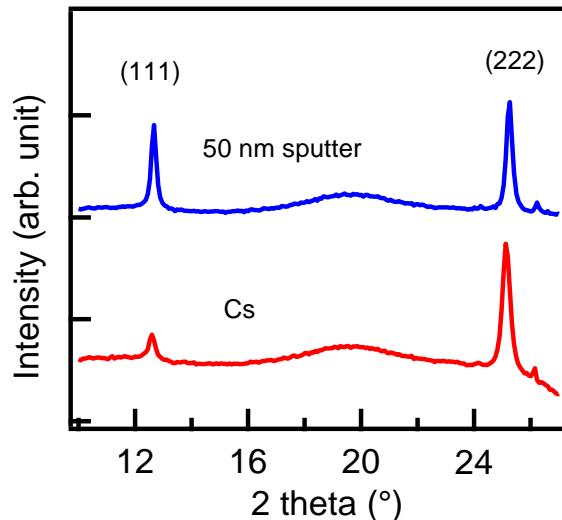
Sputter Growth

25 nm K_2CsSb + layers of (total 30 nm) Cs evap.
Silicon substrate at 90 C, layer barely crystalline

25 nm sputtered layer + Cs

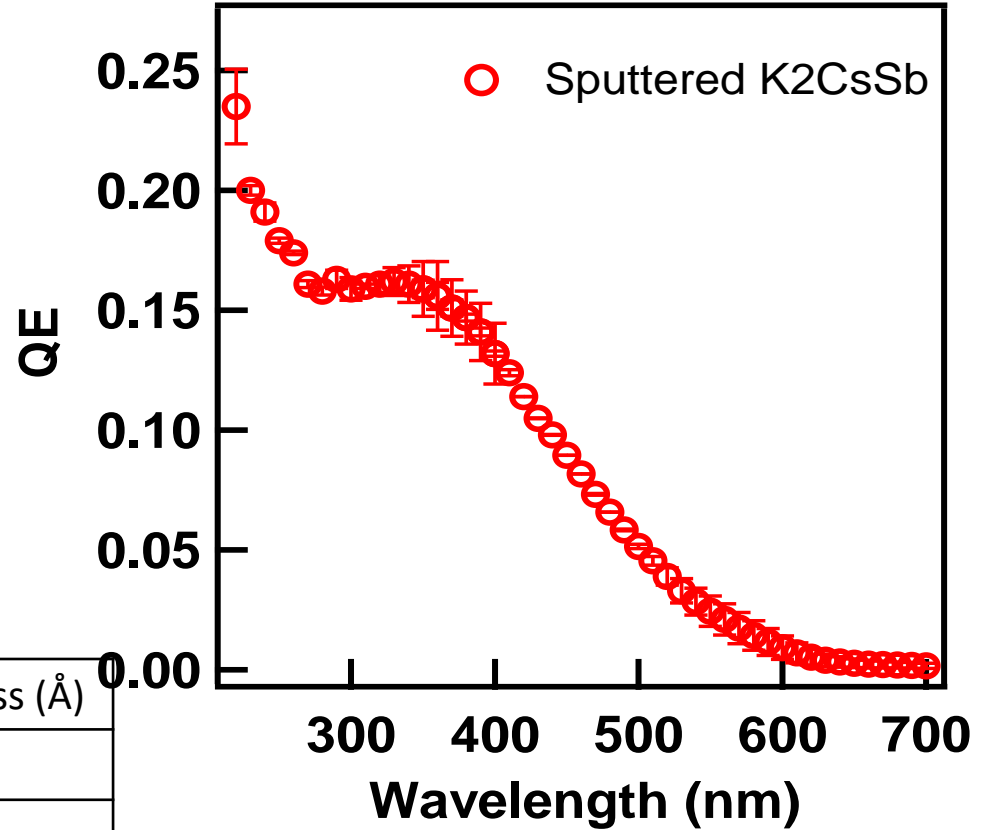
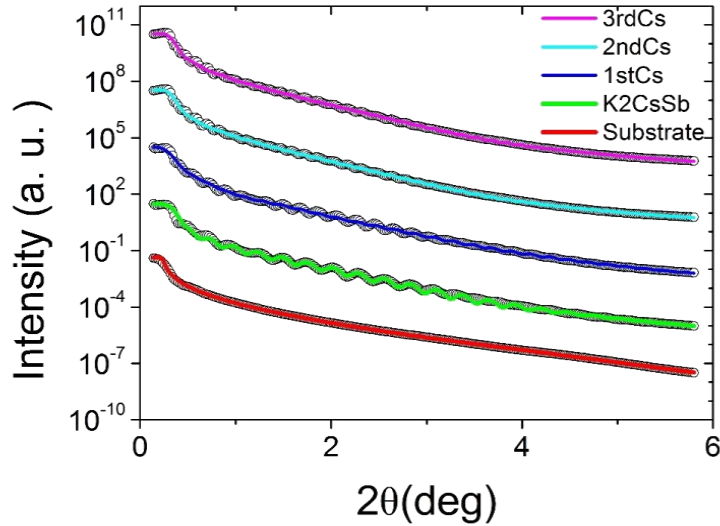


50 nm sputtered layer + Cs



layer	K (± 0.1)	Sb (± 0.05)	Cs (± 0.05)	K/Cs
K_2CsSb sputter	0.85	1.00	0.41	2.08
Cs	0.84	1.00	1.75	0.48

Surface roughness & QE of Sputtered Photocathodes

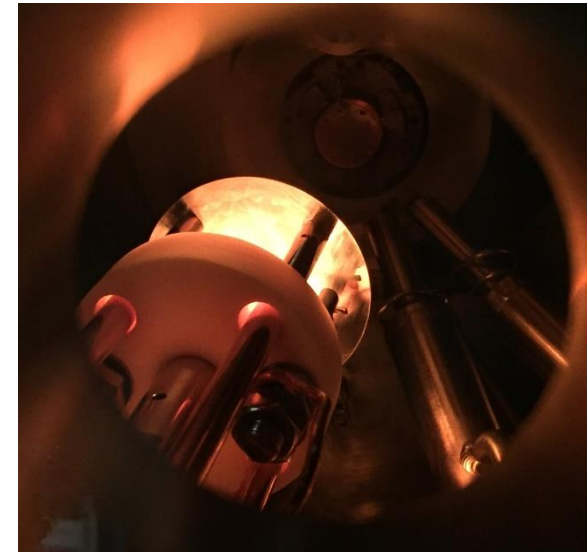
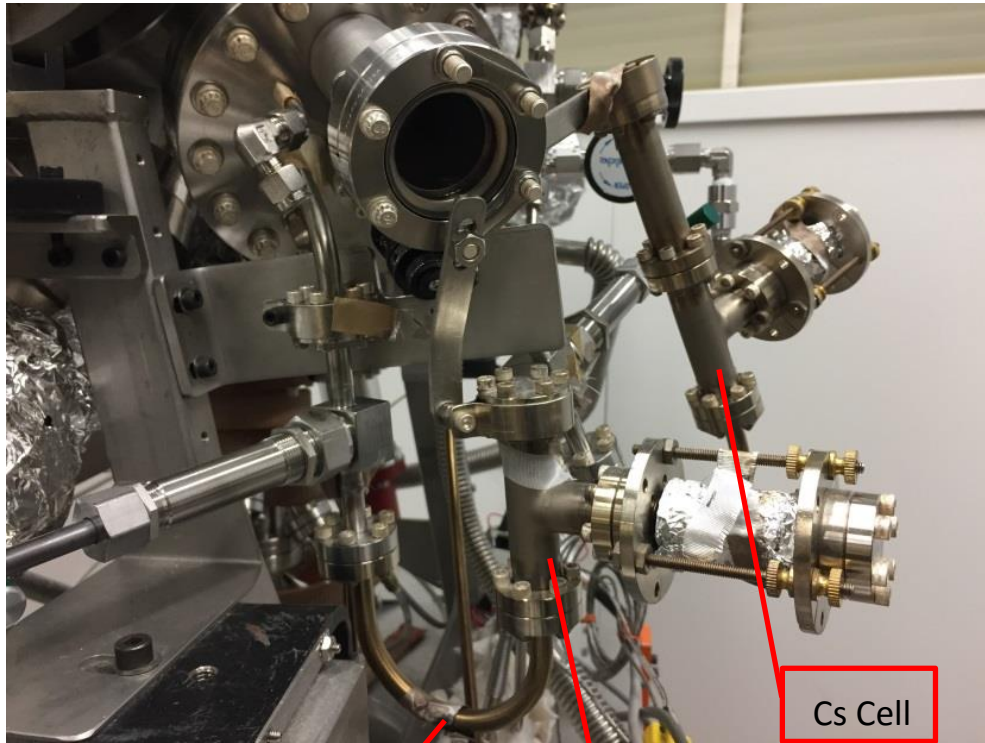


	Thickness (Å)	Roughness (Å)
3 rd Cs	416.0	5.67
2 nd Cs	341.3	4.94
1 st Cs	249.5	4.91
sputter K ₂ CsSb	234.2	5.17
SiO ₂	10.24	3.27
Substrate (Si)	---	3.75

- Peak QE > 20%
- Green QE: 4.3%

Ternary Co-evaporation

Simultaneously evaporate from Sb evaporator and K,Cs effusion cells



In situ, In operado XRR, XRF, XRD & Quantum efficiency (QE) measurement

J tube

K capsule

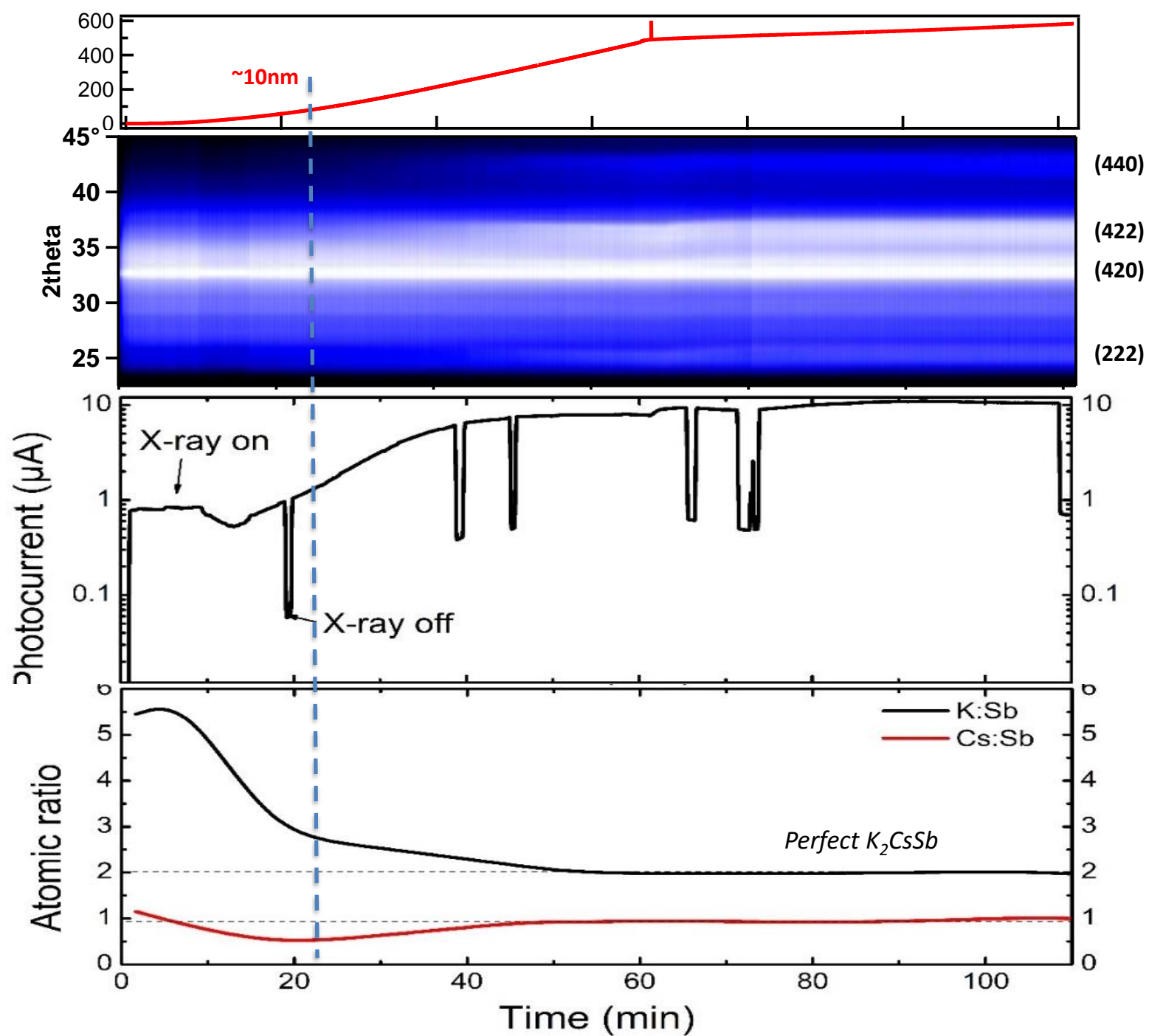
Growth rate are controlled by J tube temperature, valve and shutter

Thickness

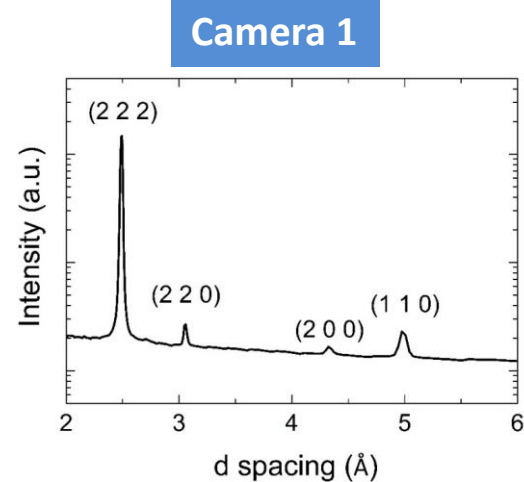
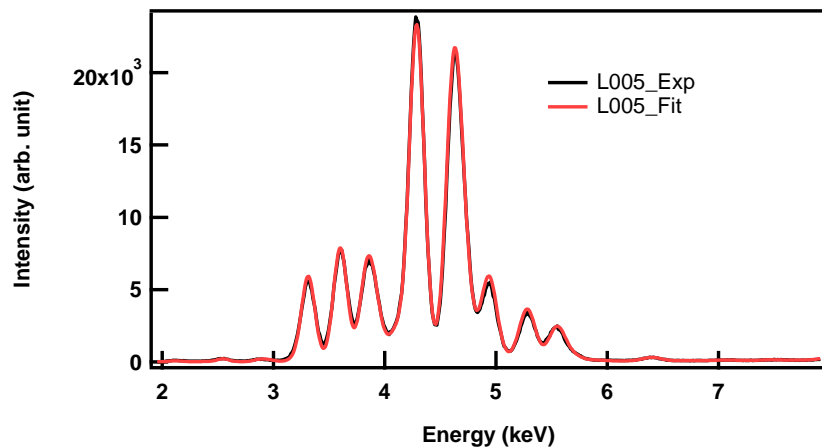
Real time
XRD

QE

Real time
Fluorescence

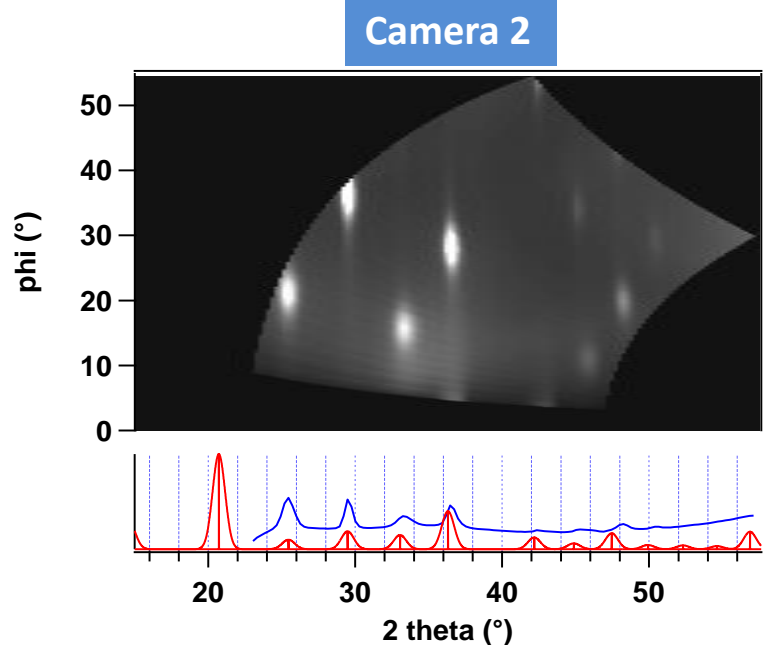


Stoichiometry & Structural Analysis



	K	Sb	Cs
L004 Si	2.50	1.00	1.16
L005 Si	2.37	1.00	0.91
L006 Si	2.21	1.00	0.95
L011 Si	2.07	1.00	0.94
L012 MgO	1.98	1.00	0.88

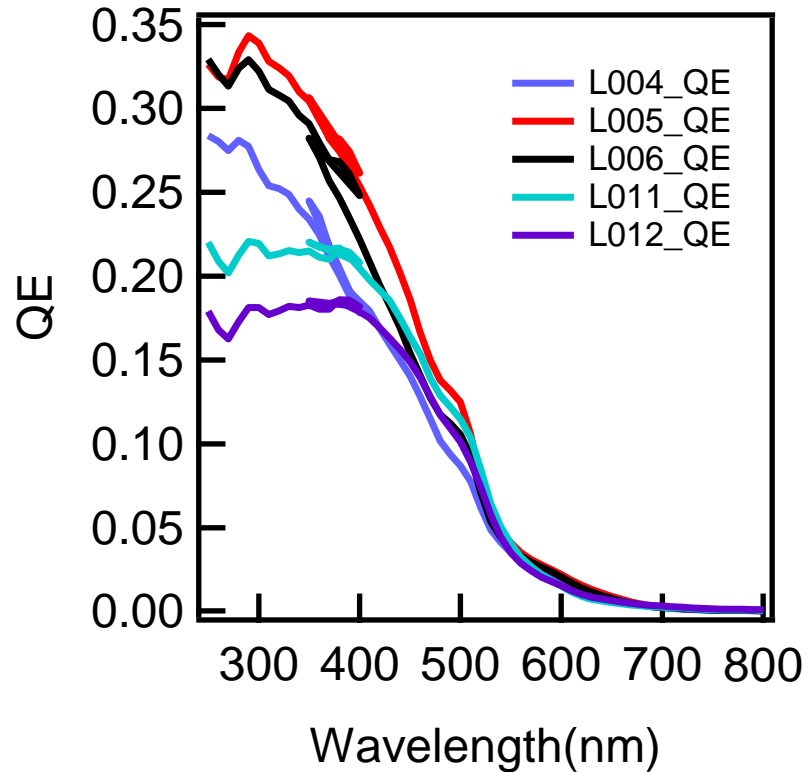
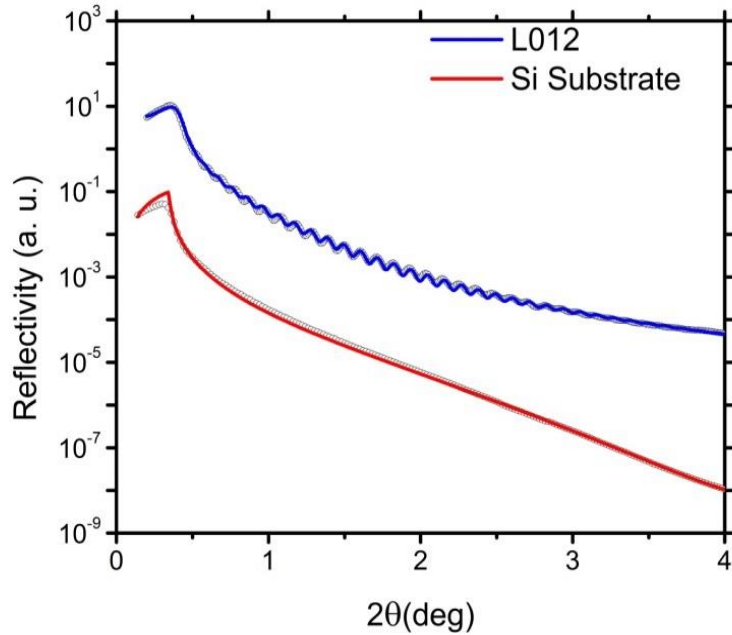
Good K/Cs/Sb ratio!



Highly textured K_2CsSb phase!

This works for the entire Alkali antimonide family – we've created pseudo single crystals with a wide range of stoichiometries

Surface Roughness & QE



	QE@532nm(%)	Roughness(A)	Thickness (A)	Grain size (A)
L004 Si	4.9	3.5	234	155
L005 Si	5.8	11.5	815.3	277
L006 Si	5.4	13.8	757.5	202

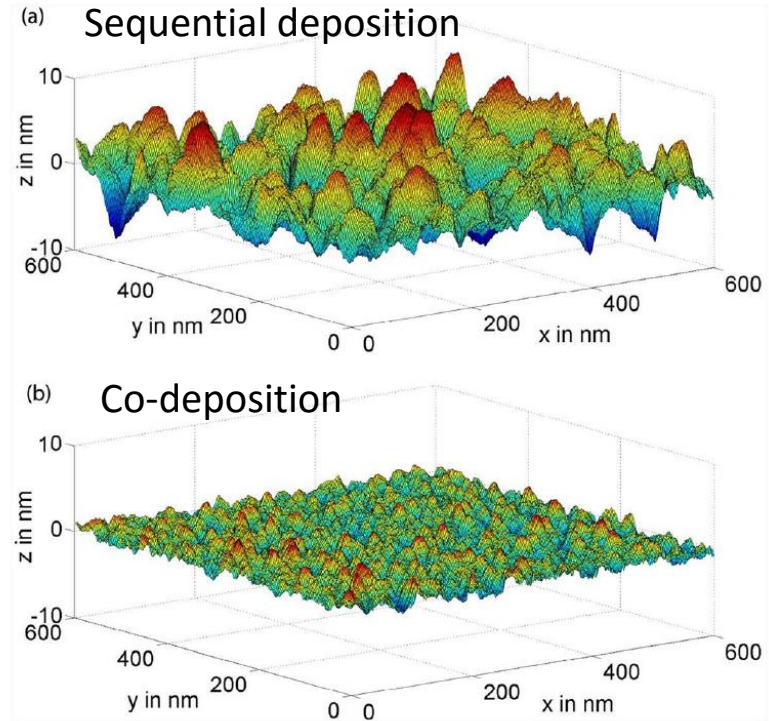
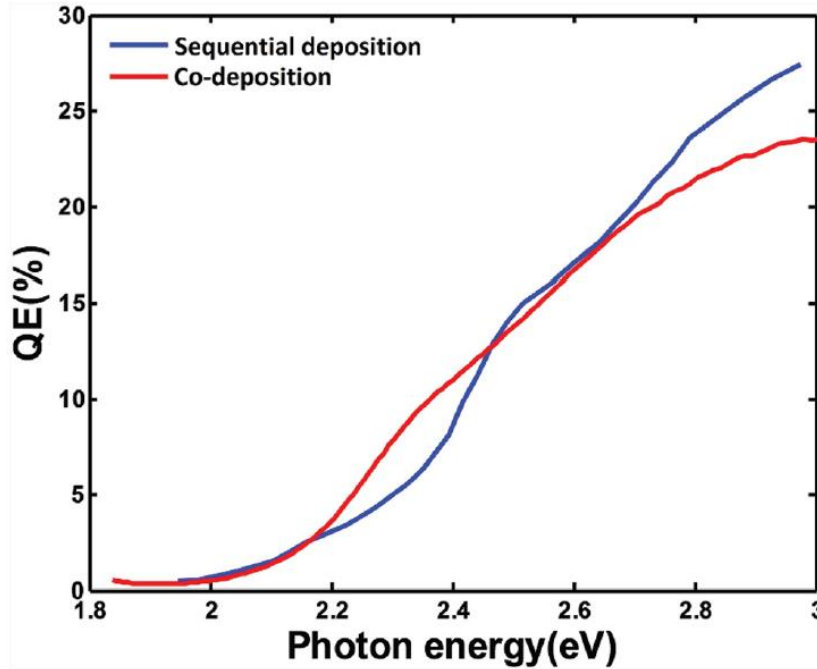
Simultaneous evaporation of all constituents results in no crystal phase transformation

Smooth, and High crystal quality => High QE

Co-deposition leads to efficient and ultra-smooth cathodes

Ternary Co-evaporation at LBNL

Co-deposited K2CsSb yield spectrum



Quantity	Sequential deposition	Co-deposition
RMS roughness (nm)	2.5	0.6
ϵ_5 ($\mu\text{m}/\text{mm}$ rms)	0.18	0.07
ϵ_{20} ($\mu\text{m}/\text{mm}$ rms)	0.36	0.14
ϵ_{100} ($\mu\text{m}/\text{mm}$ rms)	0.80	0.31

5 MV/m

20 MV/m

100 MV/m

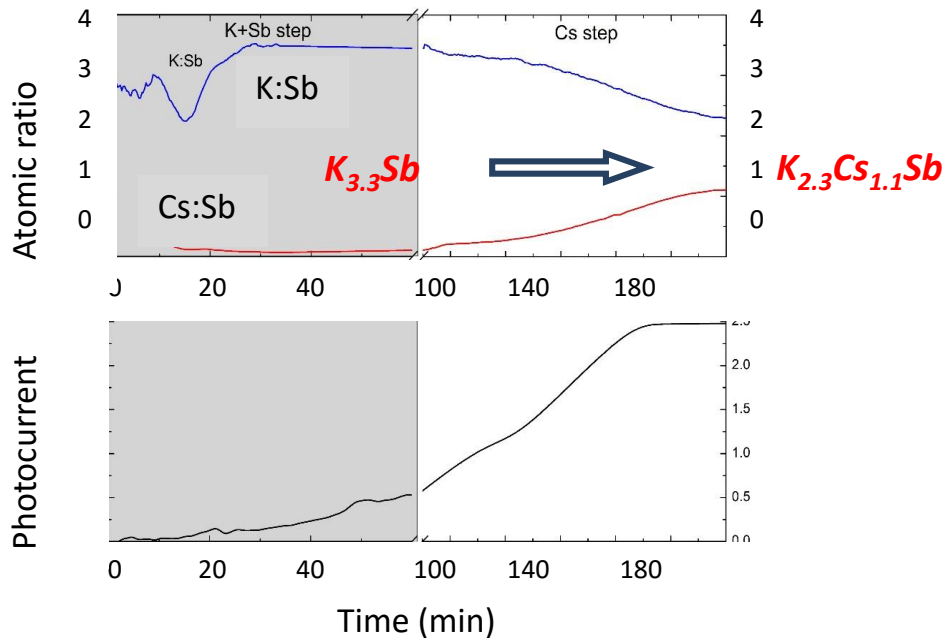
20 MV/m

- significant emittance degradation for sequential method
 - no practical degradation with co-deposition
- LN2 threshold emission
 - significant degradation, even with co-deposition
 - we need films smoother than 0.6 nm!!

2-step Co-evaporation

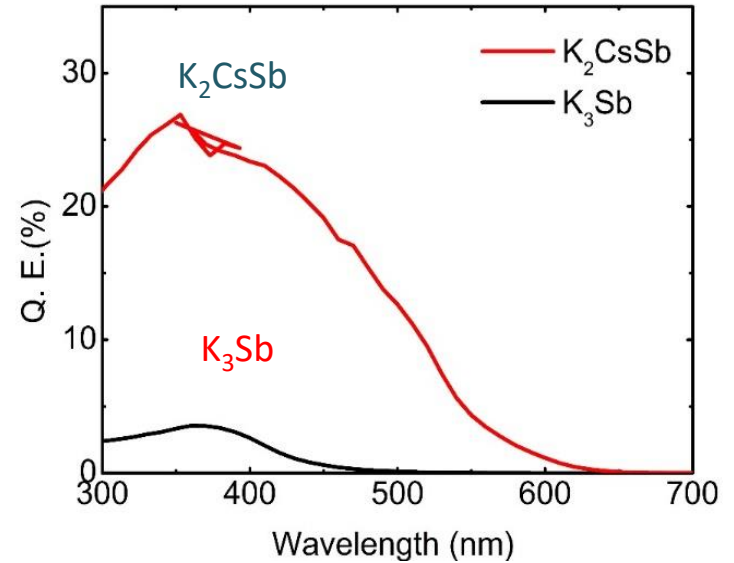
Two-step recipe:

1. K+Sb co-deposition, maximize QE at 380 nm, @ ~120°C
2. Cs deposition on top @ ~100°C until QE (530 nm) maximizes



K:Sb becomes close to 3:1 shortly after K-Sb co-deposition starts

Spectral response:



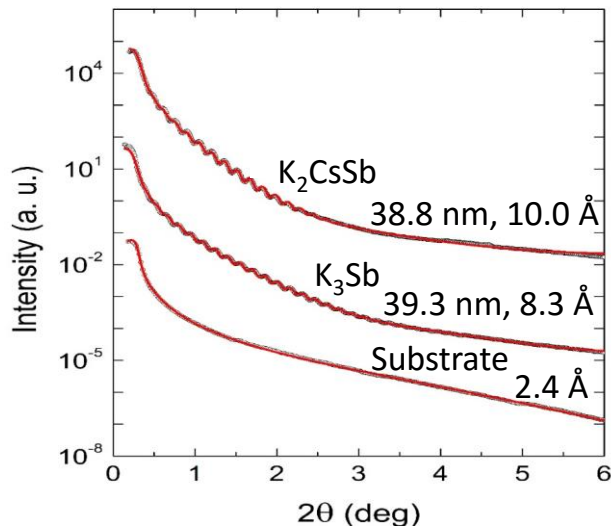
- K_3Sb layer: peak 3.5% at 360 nm; 0.047% at 530 nm
- After Cs: peak 26% at 360 nm; 7% at 530 nm

Most of the performance advantages of Ternary co-evaporation, but MUCH easier

2-step Co-evaporation

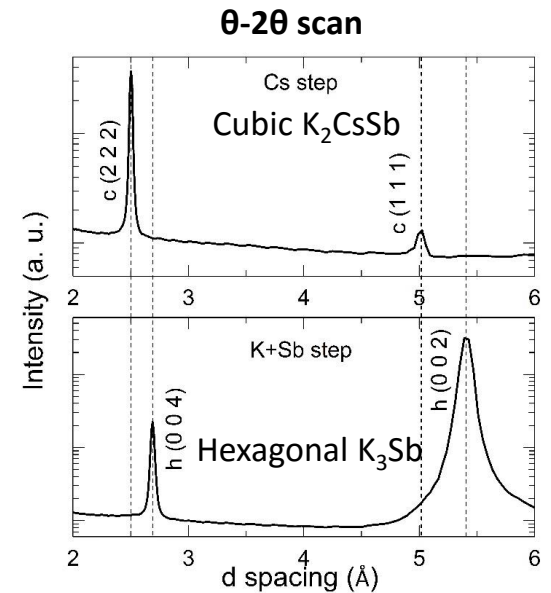
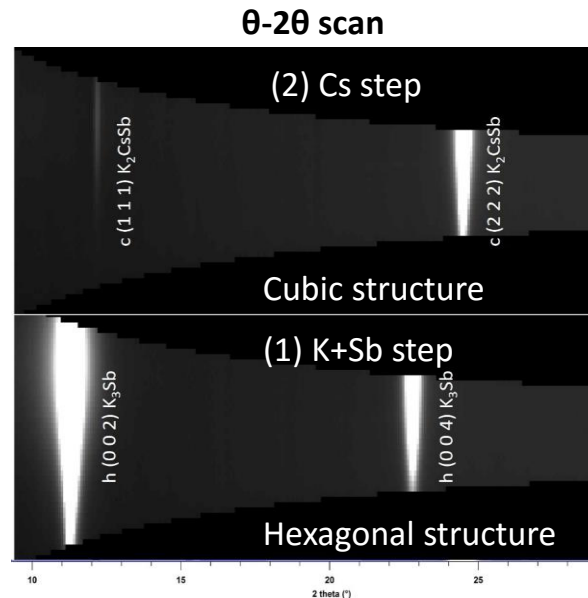
XRR

Sub-nm roughness
 Similar thickness, roughness
 from K_3Sb to K_2CsSb



XRD

Full conversion from hexagonal K_3Sb to perfect
 cubic K_2CsSb
 Textured XRD pattern

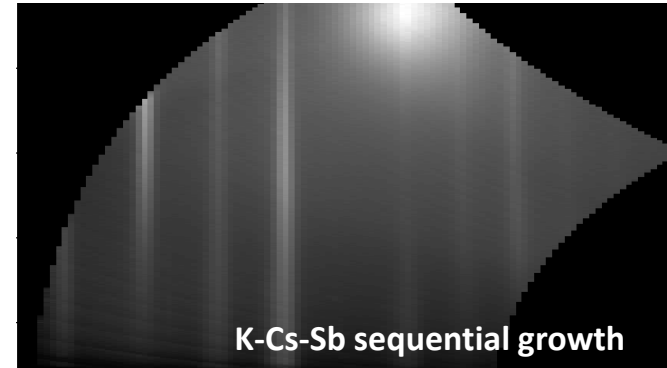
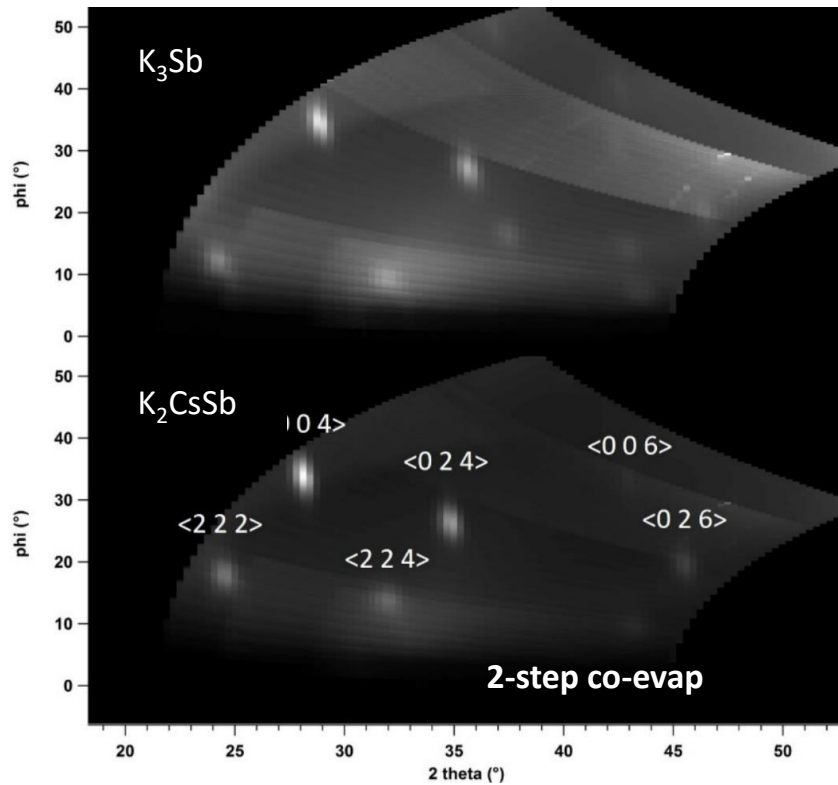


Diffraction arcs – textured film (both on Si (1 0 0) & Si (1 1 1))

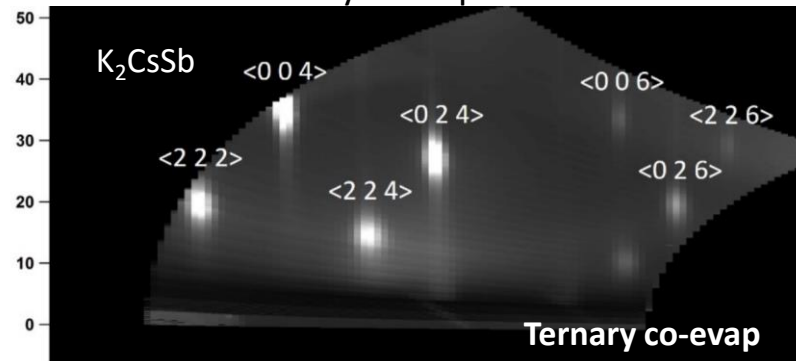
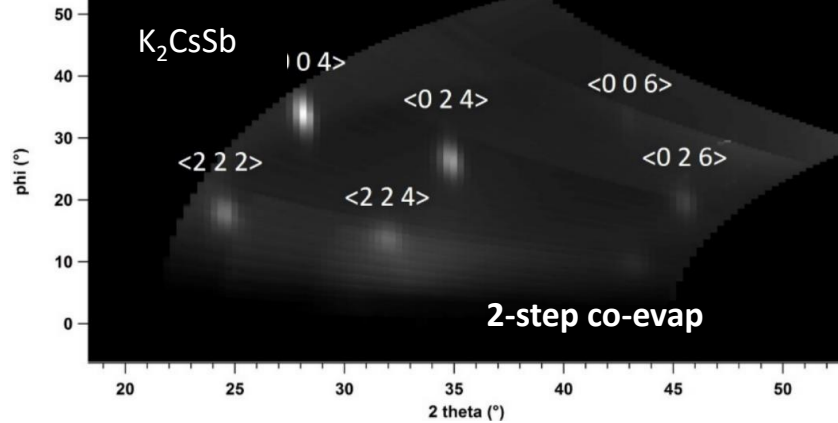
Crystal Quality

Wide angle X-ray diffraction patterns

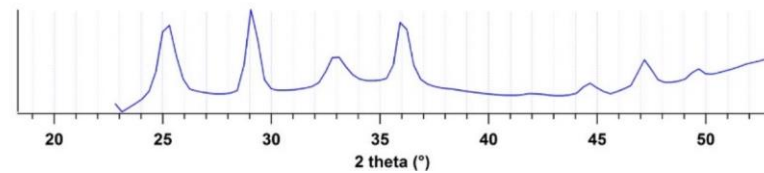
Two-step recipe



Ternary co-deposition



Large Grain, highly textured

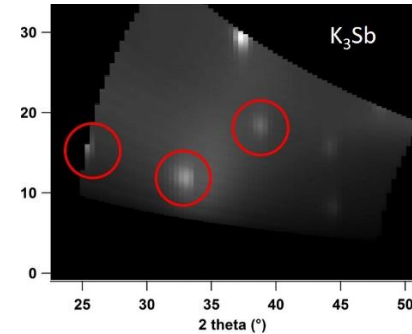
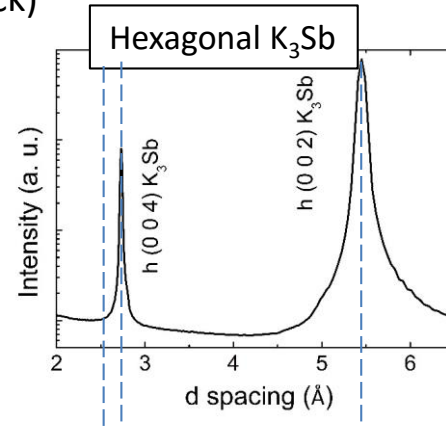
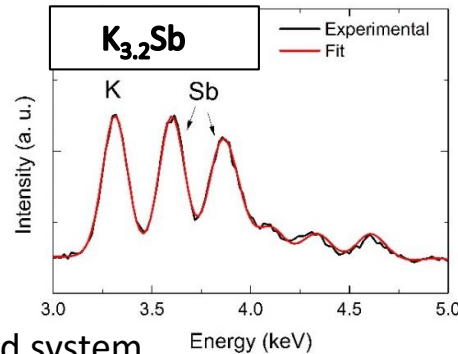


Effusion cells – bilayer photocathode

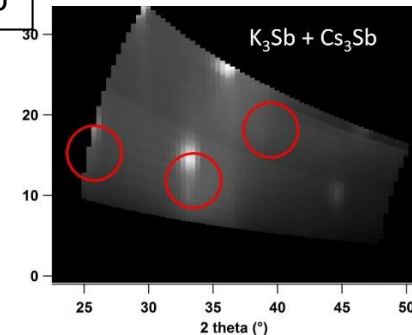
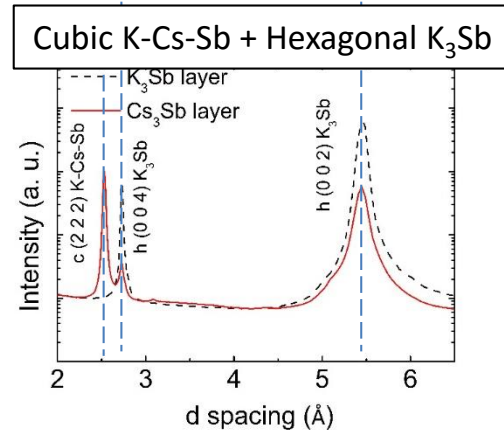
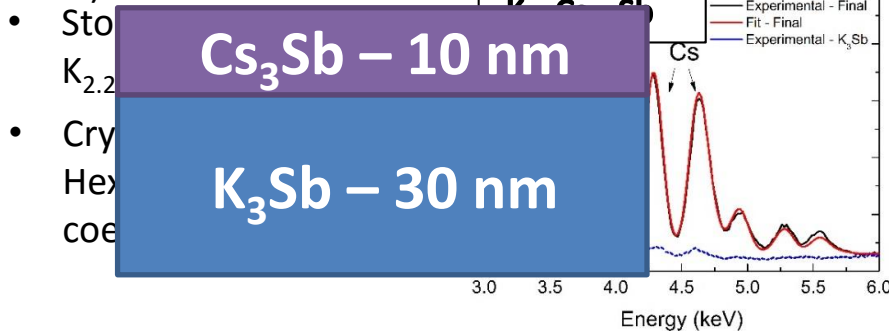
Controllable stoichiometry, crystal structure from effusion cells – *Heterojunction fabrication*

- K_3Sb layer (K-Sb co-dep at 120 °C, ~30 nm thick)

- Stoichiometry: $K_{3.2}Sb$
- Crystal structure: Hexagonal; highly textured



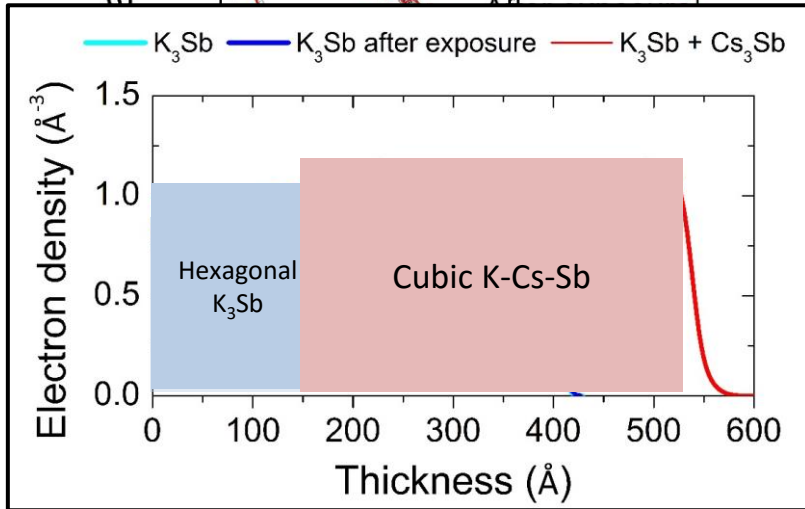
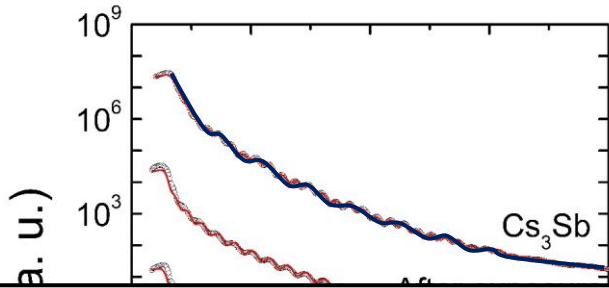
- 5 min exposure to unbaked system
- Cs_3Sb layer (Cs-Sb co-dep at RT, 10 nm thick)



Effusion cells – bilayer photocathode

- Bilayer structure

XRR simulation



Different stages during bilayer photocathode fabrication

Stage	Thickness (Å)	Roughness (Å)
After Cs ₃ Sb	543 (-7.10, 6.46)	7.89 (-0.15, 0.21)

- From K₃Sb to K₃Sb/Cs₃Sb:
XRD result: thin K₃Sb at bottom,
thick K-Cs-Sb at top

Final bilayer photocathode

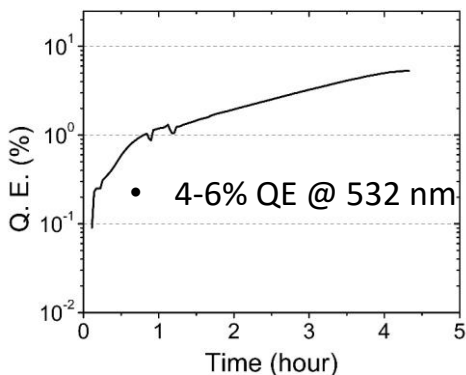
Layer	Thickness (Å)	Roughness (Å)	Electron density (Å ⁻³)
K-Cs-Sb	373 (-2.0, 1.2)	7.89 (-0.06, 0.08)	1.16
Interface	7.86 (-3.55, 4.20)	7.10 (-0.52, 1.14)	1.03
K ₃ Sb	154 (-3.5, 3.9)	30.3 (-2.3, 1.3)	0.99
Substrate	-	2.32	0.72

Not the heterojunction we were trying for... but still a sharp heterojunction

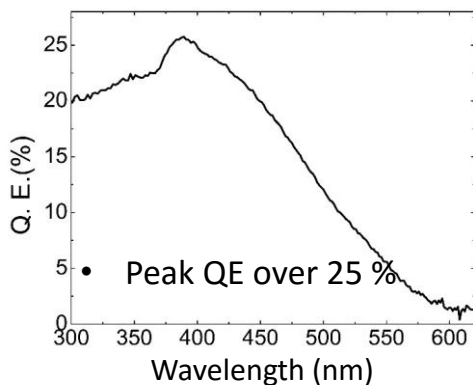
K₂CsSb thermal decomposition

Photocathode growth and lifetime study at Berkeley Lab

Typical QE increase curve during K-Cs-Sb co-dep growth

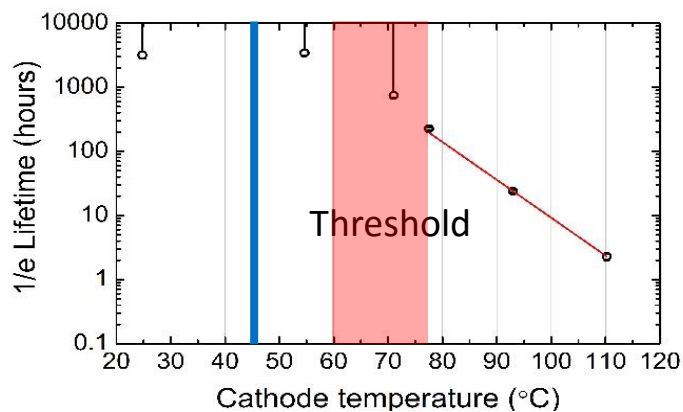


Typical spectral response of K-Cs-Sb photocathode



Lifetime at elevated cathode temperature

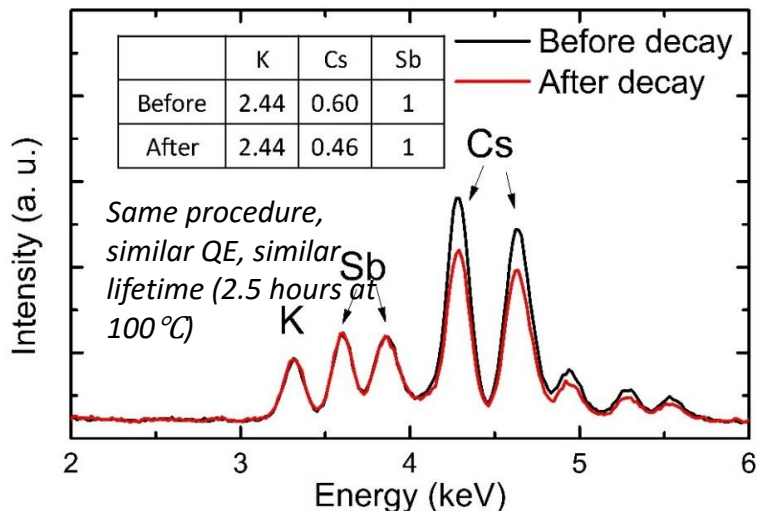
Temperature (°C)	Exponential fit – first term		Exponential fit – second term	
	Decay constant (τ_1)	Amplitude (A_1)	Decay constant (τ_2)	Amplitude (A_2)
RT	3167	1.06	-	-
55	3443	1.08	-	-
71	741	1.05	-	-
77	761.9	0.51	15.7	0.52
93	114.3	0.39	7.63	0.61
110	28.8	0.18	1.65	0.85



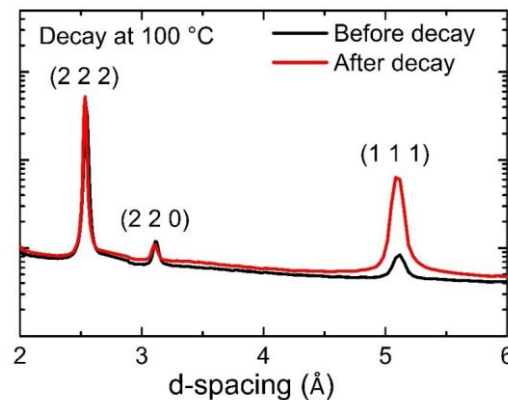
K₂CsSb thermal decomposition

X-ray characterization on co-deposited K-Cs-Sb at CHESS, decayed at 100°C

X-ray fluorescence



X-ray diffraction



Peak shift
0.34% lattice constant decrease

A small amount of Cs leaves the lattice

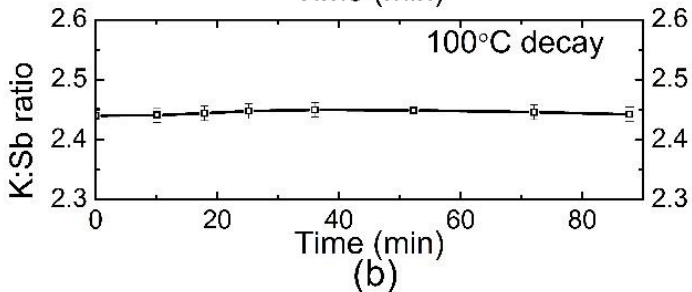
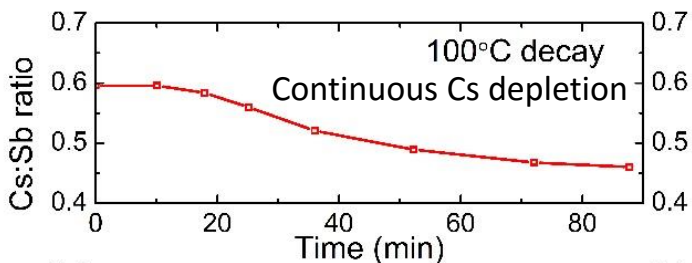
X-ray reflectivity

Layer	Thickness (Å)	Roughness (Å)
After decay	516	13.9
	(-4.49, 3.81)	(-0.10, 0.25)
Before decay	541	11.5
	(-2.76, 4.02)	(-0.24, 0.63)
Substrate	-	2.35
		(-0.01, 0.02)

5% thickness decrease

Cs leaves the surface/grain boundaries

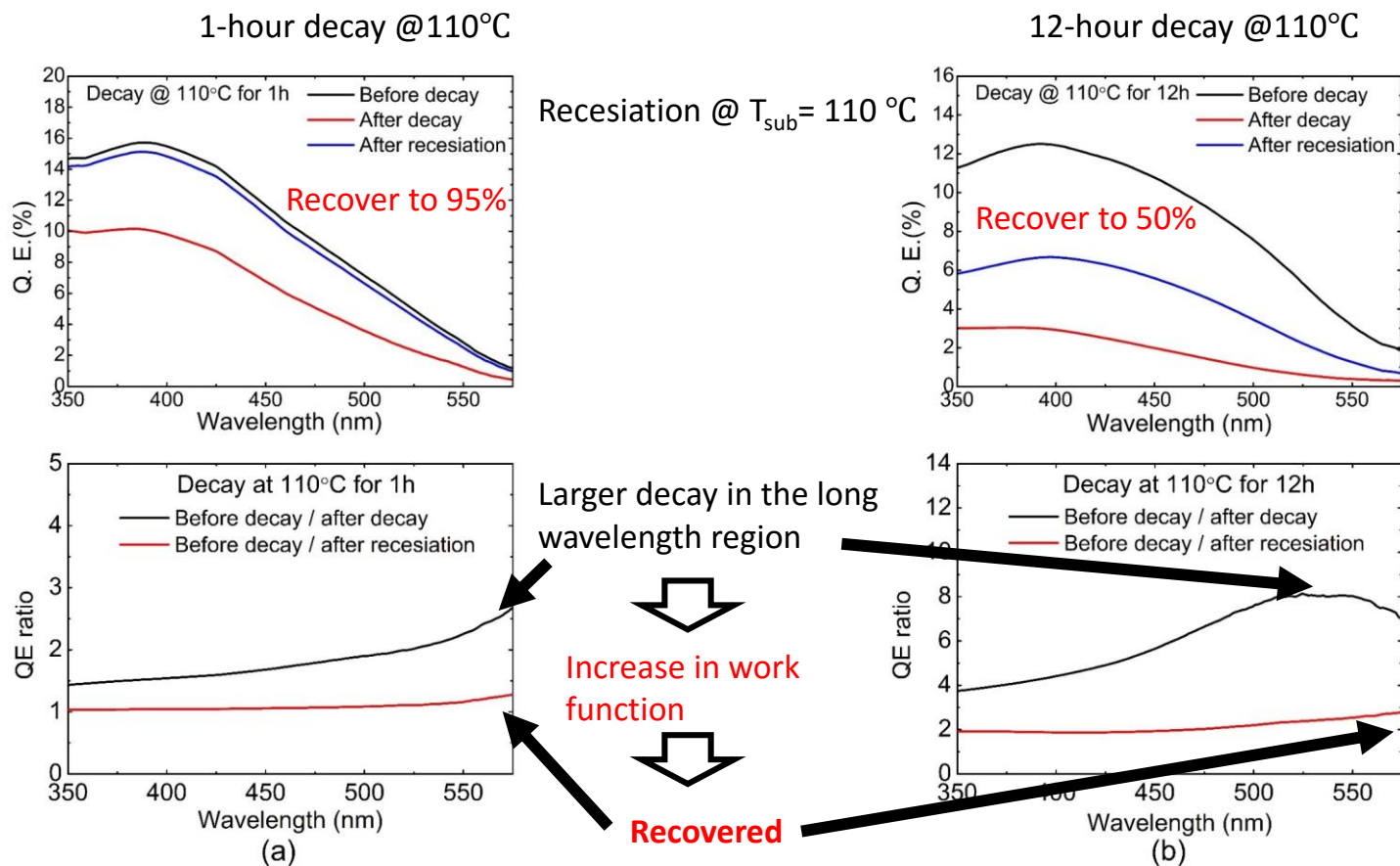
Major cause



(b)

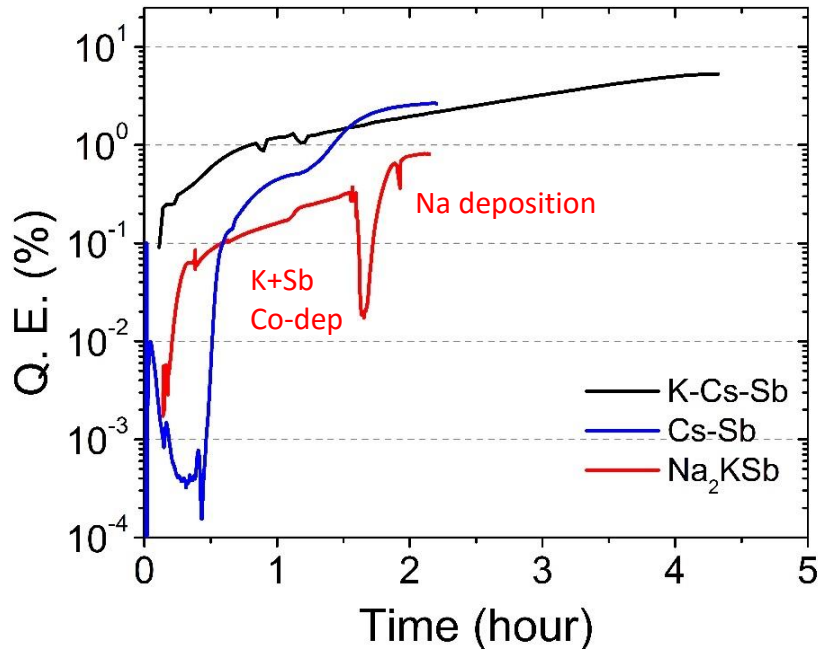
K₂CsSb thermal decomposition

Spectral response measurement and recesiation at Berkeley Lab



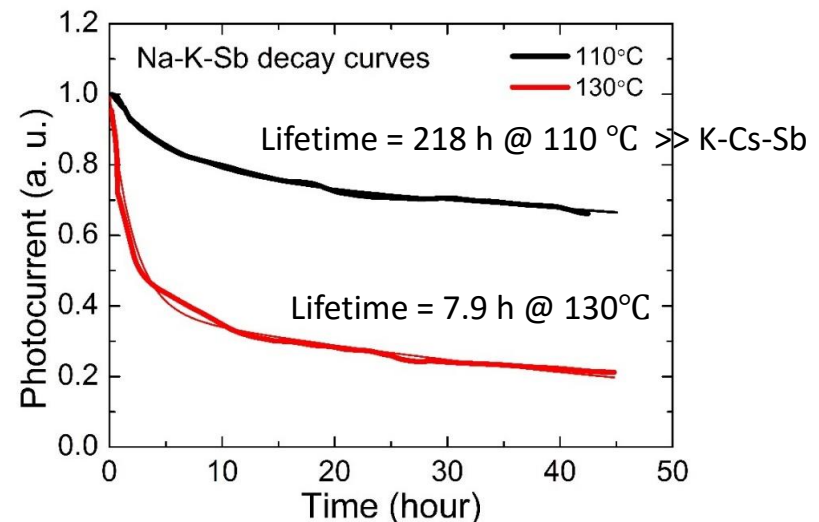
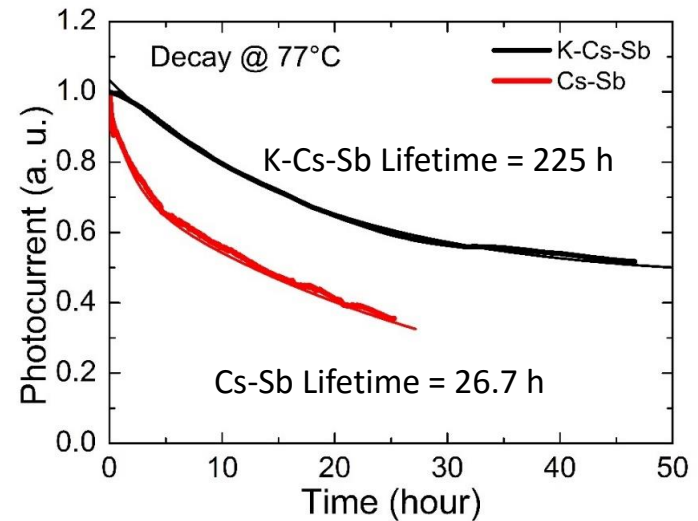
Alkali Antimonide thermal decomposition

Comparison with other alkali antimonide
Co-deposited Cs-Sb and sequential Na-K-Sb



Robustness (lifetime at higher T_{sample}):

Na-K-Sb > K-Cs-Sb > Cs-Sb



Alkali Antimonide Cathodes

What we've learned

- We now have a tool which is capable of optimizing growth parameters for figures of merit other than Quantum Efficiency, and to specifically target material properties.
- We understand the formation chemistry of these materials, and why traditional deposition results in rough cathodes
- RMS roughness down almost 2 orders of magnitude, to ~atomic scale
- Avoiding crystalline Sb helps, as does co-evaporating alkali
- Sputter deposition is good – easy to do, covers large area, almost atomically smooth even for thick films – but alkali poor
- Real time XRF feedback provides option of ternary coevaporation, producing best cathode
- Can now consider heterojunctions and doping of alkali antimonides (following a similar development path to the III-V materials)
- Can consider ultra thin (under 10 nm) cathodes to improve response time
- Conformal coating of structured surfaces possible

Thanks for your attention!

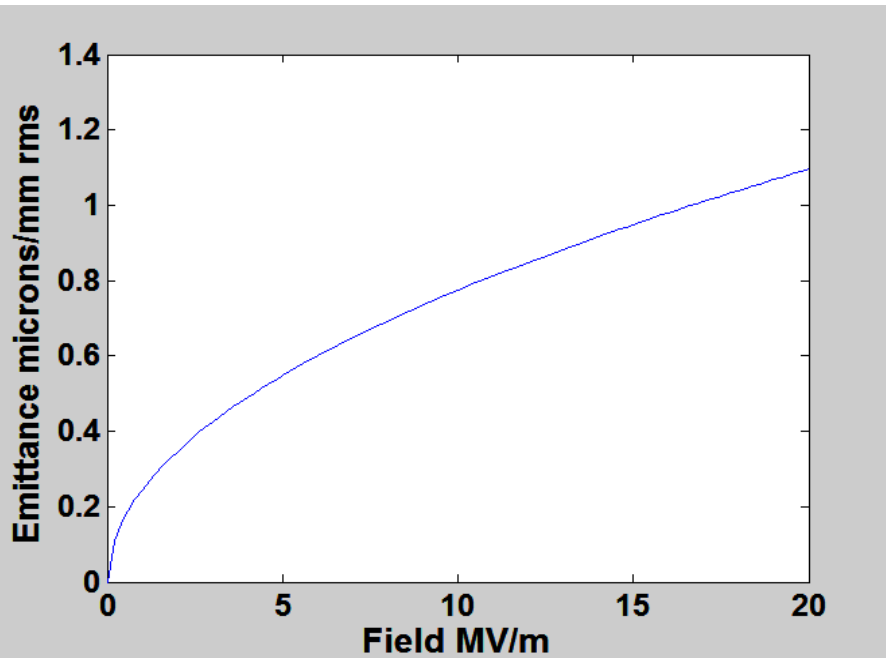
- Thanks to K. Attenkofer, S. Schubert, M. Ruiz Oses, J. Xie, J. Kuhn, M. Schmeißer, J. Wang, H. Padmore, E. M. Muller, M. Gaowei, J. Walsh, T. Vecchione, J. Sinsheimer, Z. Ding
- DOE Office of Science – Basic Energy Science and Nuclear Physics



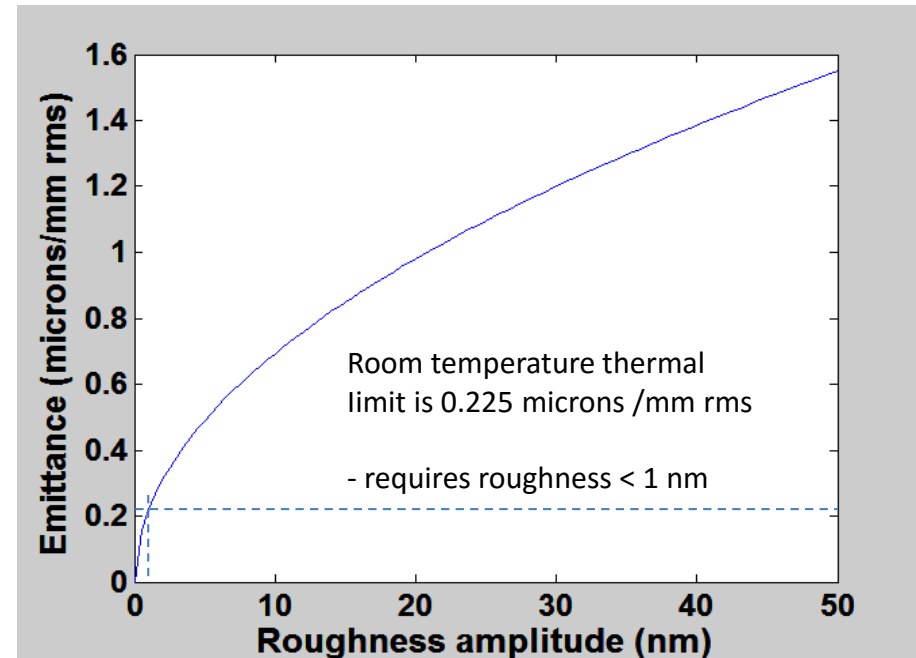
Roughness and Emittance

$$\mathcal{E}_{rough} = \sigma_{x,y} \sqrt{\frac{\pi^2 a^2}{2m_0 c^2 \lambda}} Ee$$

D. Xiang et al. Proceedings of PAC07, Albuquerque, New Mexico, USA



Field dependent emittance growth: 20 nm amplitude, 80 nm period



Emittance growth at 20 MV/m, period 4 x amplitude.

Calculations by H. Padmore